Alkaloids

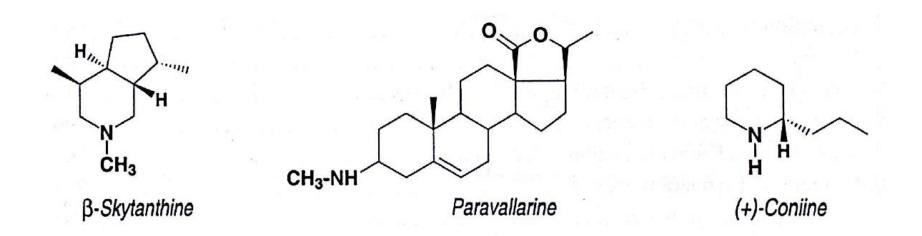
Dr. Kursinszki László

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Alkaloid is an organic compound of natural origin,

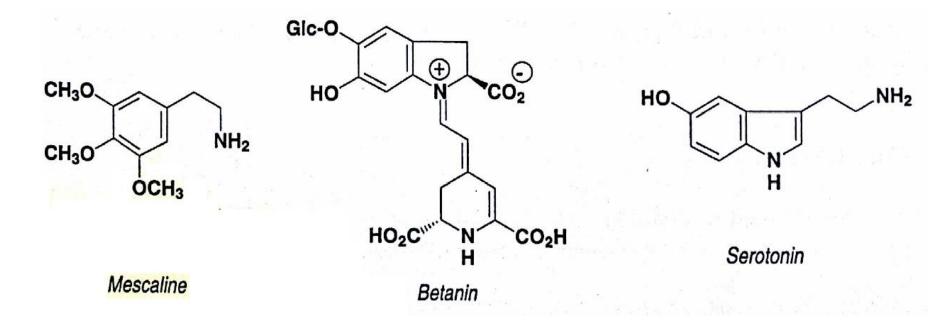
- which contains a *nitrogen atom*, is more or less *basic*,
- is of *limited distribution*, and
- has, at low doses, marked pharmacological properties.
- These compounds have in common some reactions of precipitation with the "general reagents for alkaloids".

- **Pseudoalkaloids** most often have all of the characteristics of the true alkaloids, but they are **not derived from amino acids**.
 - Terpenoid alkaloids: monoterpenoids (e.g., β-skythanine), sesquiterpenoids (from the Nymphaceae), diterpenoids (e.g., aconitine), steroidal alkaloids (e.g., paravallarine).
 - Heterocyclic nitrogen-containing substances arising from the metabolism of acetate (e.g., coniine, the toxic priciple of hemlock).

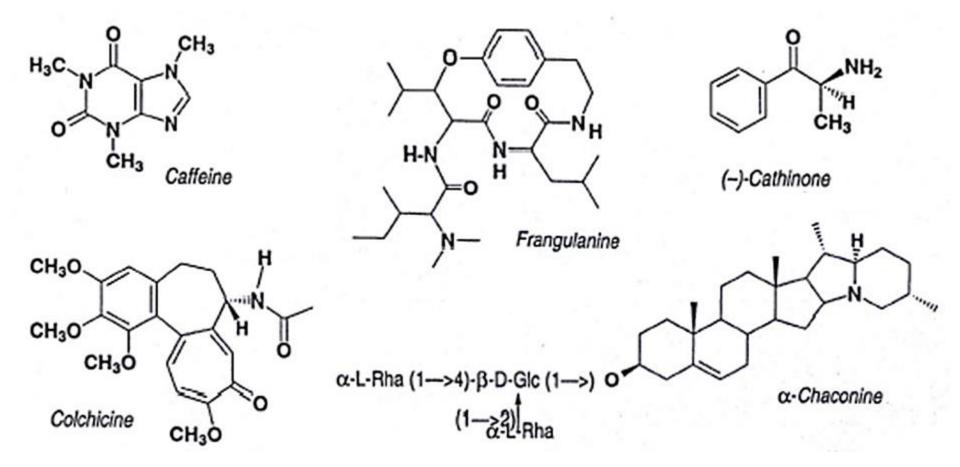


Protoalkaloids are *simple amins* in which **the nitrogen atom is not part of a heterocyclic ring**; they *are basic* and are *elaborated in vivo from amino acids*.

- Various substances fullfil this definition:
 - Simple amins, such as serotonin, mescaline from peyote, or cathinone from Abessiniantea,
 - **Betains** (resulting from the quaternization of of the nitrogen atom of amino acids)
 - Some authors include **betalains** (*"chromoalkaloids"*) in this group (e.g., Betanin).



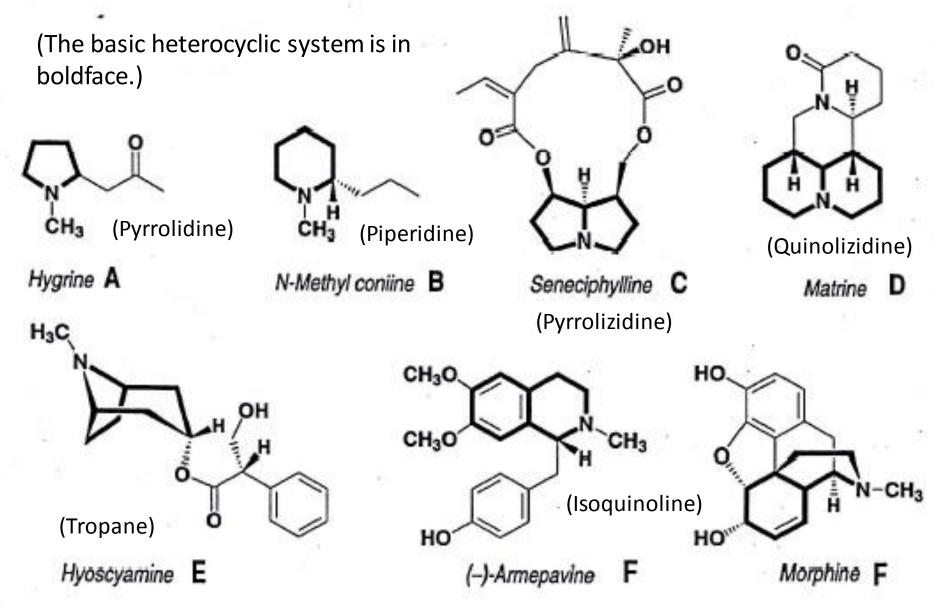
Athough the distinction between true alkaloids, protoalkaloids, and pseudoalkaloids is intellectually appealing, it is not always easy to apply.



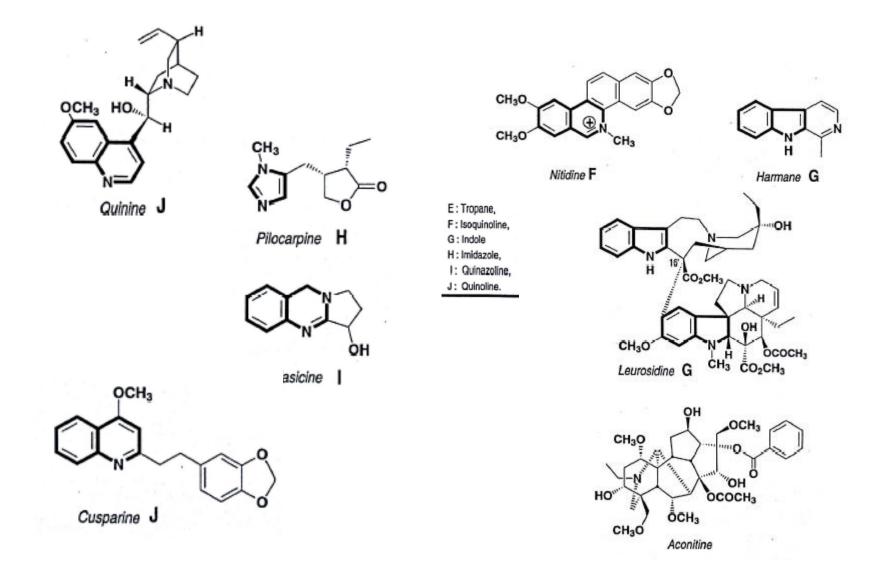
In practice, it is widely accepted that the folloving are not alkaloids:

Simple amins, peptides, amino sugars, porphyrins, alkylamins, and arylamins, at least those that are widely distributed.

Examples of alkaloid structuters illustrating the chief heterocyclic system encountered.

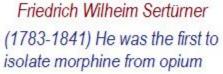


Examples of alkaloid structuters illustrating the chief heterocyclic system encountered.



History









Pierre Joseph Pelletier Joseph Bienaimé Caventou (1788-1842) (1795-1877) Isolation of strychnine (1818), brucine (1819), quinine (1820) and caffeine (1821)



Kabay János (1896-1936) He earned morphine from the dry poppy-straw. (1925)



Sir Robert Robinson Nobel Prize in 1947 for his research on plant dyestutts (anthocyanins) and alkaloids



Robert Burns Woodward Nobel Prize in 1965 The total synthesis of natural products, e.g. strychnine

ALKALOIDS: PHYSICO-CHEMICAL PROPERTIES

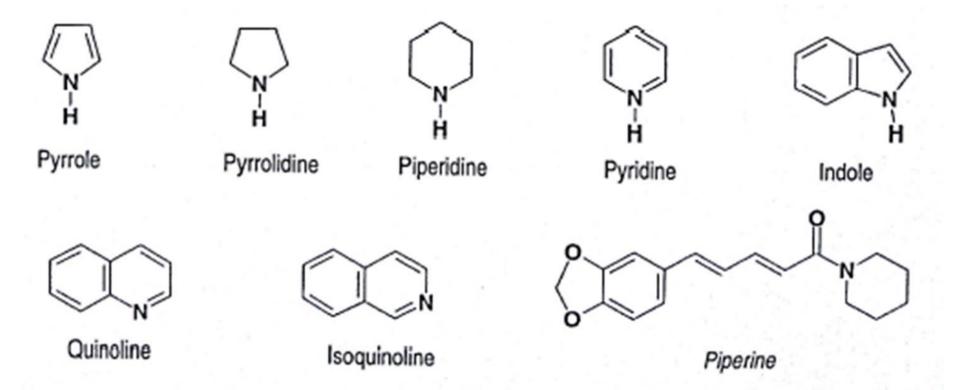
- Alkaloids have **molecular weights** ranging from 100 to 900.
- **Oxigen-free bases** generally are **liquid** at ordinary temperatures (nicotine sparteine, coniine).
- **Bases containing oxygen atoms** are normally **crystallizable solids**, some of them are **colored** (berberine).
- Crystallized bases generally
 - rotate the plane of polarized light, and
 - have **sharp melting points**, without decomposition, especially below 200 °C
- Solubility. As a general rule, alkaloids as bases are
 - not soluble or are sparingly soluble in water,
 - soluble in apolar or only slightly polar organic solvents, and concentrated hydroalcoholic solutions.

ALKALOIDS, BASICITY

- Basicity varies greatly.
- Depends entirely on the availability of the lone pair of electrons on the nitrogen atom.
- In close proximity to the nitrogen atom, electron withdrawing groups ↓, electron donating groups ↑ the basicity. →
- **Colchicine** and **piperine are**, because of the presence of the carbonyl group on the amide, practically **neutral**.
- The basic character of the heterocyclic ring itself varies depending on that the lone pair of electrons on the nitrogen atom is available or plays a role in the aromatic character (see next slide).
- The basicity is also influenced by **steric constraints.**

- As bases
 - in solution they are sensitive to heat, light, and oxygen;
 - they form salts with mineral acids (hydrochlorides, sulfates, nitrates), or organic acids (tartrates, sulfamates, maleates).
- Alkaloid salts are generally soluble in water and in dilute alcohols, and they are, except in rare cases, not soluble in organic solvents.

ALKALOIDS, The basic character of the heterocyclic ring



- The basic character of the heterocyclic ring itself varies depending on that **the lone pair of electrons on the nitrogen atom is available or plays a role in the aromatic character.**
- Quinoline, isoquinoline: bases; Pyrrole, indole: acids, Pyrrolidine: saturated, strong bases.

ALKALOIDS: DETECTION AND CHARACTERIZATION I.

General reactions of precipitation :

- Preliminary extraction can be a "classic" alkaloid extraction or alcoholic maceration.
- They are based on the fact that alkaloids form combinations with metals and metalloids: bismuth, mercury, tungsten, and iodine.
- The "general reagents for alkaloids" are used:
 - solutions containing iodine and iodide (), potassium iodide and mercuric chloride (Mayer's reagent),
 - bismuth nitrate and potassium iodide (Dragendorff's r.).
 - It is also possible to use silicotungstic acid (a mixture of tungsten and silicon oxides), or alkaline solutions of iodoplatinates.
- The specificity of these reagents is nor absolute.

ALKALOIDS: DETECTION AND CHARACTERIZATION II.

Color reactions characteristic of subgroups of alkaloids:

- *p*-dimethylaminobenzaldehyde for the ergot alkaloids and pyrrolizidine alkaloids;
- **Cerium and ammonium sulfate**, which differenciate indoles (yellow), dihydroindoles (red), b-anilinoacrylates (blue), oxindoles;
- **ninhydrin** for arylalkylamines;
- **the Vitali-Morin reaction** for the esters of tropic acid;
- Reagents containing ferri chloride in the presence of hydrochloric acid (tropolones) or perchloric acid (Rauwolfia).

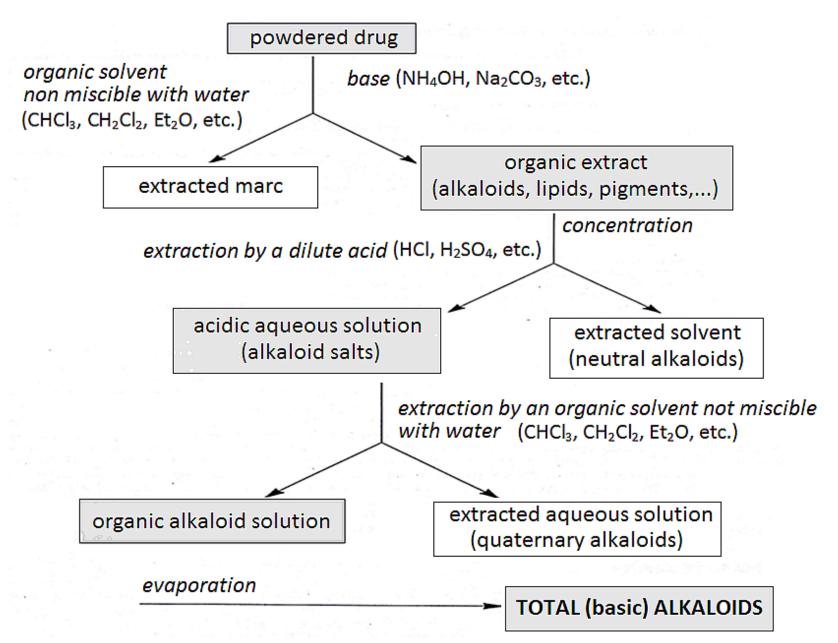
Analysis of alkaloid composition:

- Methods currently used are **TLC**, **HPLC** on normal or reversed phase, and **LC-MS**.
- Dragendorff's reagent, the iodine-iodide solution, potassium iodoplatinate, or cerium and ammonium sulfate are commonly used to visualize TLC plates.

EXTRACTION OF ALKALOIDS

- It is based, as a general rule,
- on the fact that alkaloids normally occur in the plant as salts, and
- on the differential solubility of the bases and salts in water and organic solvents.
- The **plant material often contains** substantial quantities of fats, and also waxes, terpenes, pigments, and other **lipophilic substances**, which may interfere with the extraction procedure, for example, by **causing the formation of emulsion**.
- **Preliminary defatting** of the crushed drug can solve this technical problem.
- **Petroleum ether** and **hexane** are well suited for this step: alkaloids are soluble in these solvents only in exceptional cases, when the medium is neutral.

SOLVENT EXTRACTION IN ALKALINE MEDIUM



EXTRACTION IN ACIDIC MEDIUM

Two approaches are possible: the pulverized drug is extracted with

- a., acidified water
- b., an acidified alcoholic or hydroalcoholic solution.
- In the latter case, the extraction is followed by a distillation under vacuum which eliminates the alcohol and leaves behind an acidic aqueous solution of alkaloid salts.
- In both cases, the result is an aqueous solution of alkaloid salts requiring purification. This can be accomplished by
- 1. alkalinizing the solution and extracting the bases with an immiscible organic solvent, which leads back to the above step;
- 2. selectively adsorbing the alkaloids contained in the solution on an ionexchange resin, then eluting them with a strong acid;
- **3.** precipitating the alkaloids as iodomercurates. The resulting complex is recovered by filtration, dissolved in a mixture of water, alcohol, and acetone, and decomposed by passing it through an ion-exchange resin. This technique can be used to extract quaternary ammonium salts.

ALKALOIDS, QUANTITATION (I)

Total alkaloids

- It requires preliminary extraction of the alkaloids using a general method: generally the alkaline medium approach is preferred; at each step the completeness of the extraction must be verified.
- **Gravimetric methods:** are easy to implement, but lack precision.
- Volumetric methods. Acidimetry: direct, or, most often, back titration (*p*Ka 5 -10), or in non-aqueous medium (weak bases).

Determination of **alkaloid composition**

- The available techniques include **spectrophotometry, colorimetry, fluorimetry, and densitometry**.
- Spectrophotometry: to quantitate quinine- and cinchonine-type alkaloids in Cinchona bark (Ph. Eur.)
- Colorimetry: to the quantitation of the weak bases alkaloids of Ruwolfia.
- Densitometry: TLC isolation of morfine and measurement of the reflectance directly on the plate (laboratory practice).
- **HPLC-UV, LC-MS:** tend to advantageously replace the *"*classic" methods. HPLC-UV tend to be more and more important technique for the European Pharmacopoeia.

ALKALOIDS, BIOSYNTHETIC ORIGIN

- **The precursor is**, for true alkaloids, **an amino acid**: ornithine, lysine, phenylalanine, tyrosine, tryptophan, hisitdine, or anthranilic acid.
- The formation of the alkaloid may require the involvment of only one molecule of amino acid (hygrine, cathine), or two molecules of the same amino acid (quinolizidins, benzylisoquinolines), or, less commonly, of two different amino acids (tubulosine), or else of several molecules of the same acid (sparteine).

Mechanism of the formation of the heterocyclic system:

 Generally simple inter- or intramolecular reactions : formation of a Schiff base, or, Mannich reaction...

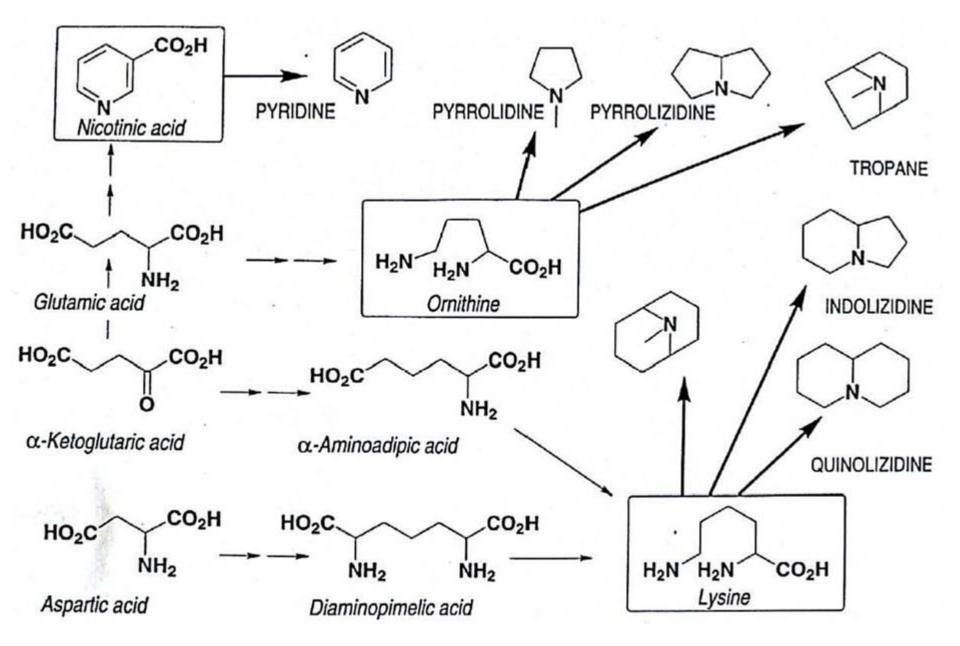
Additional carbon atoms. These come from

- intermediates that have major role in other metabolic pathways: acetate (tropane), dimethylallylpyrophosphate (ergolines, furoquinolines), or
- *intermediates more specific to* a particular group of plants, like secologanin (monoterpenoid indol alkaloids).

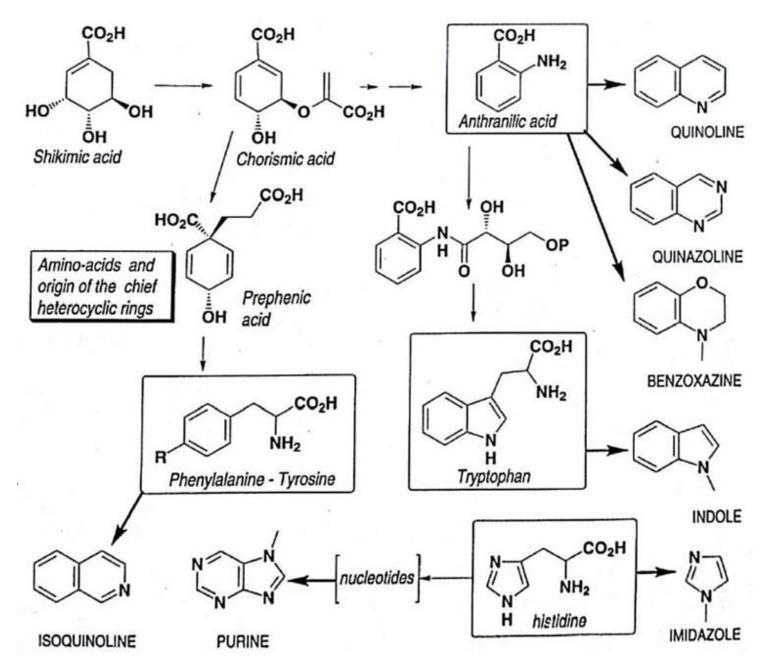
Origin of the wide structural variability: allylic oxidation, oxidative coupling, oxidation of the aromatic rings, esterifications, and etherifications.

Terpenoid alkaloids: the precursors are stricly of terpenoid origin and the formation of an amine function occurs late in the pathway.

Amino acids and origin of the chief heterocyclic rings I.



Amino acids and origin of the chief heterocyclic rings II.



ALKALOIDS, Pharmacological Activity

Alkaloids are particularly interesting substances because of their **multiple pharmacological activities**:

- on the CNS: alkaloids are depressants (morphine, scopolamine) or stimulants (strichnine, caffeine);
- on the automathic nervous system:
 - sympathomymetics (ephedrine), or
 - sympatholytics (yohimbine, certain ergot alkaloids),
 - parasympathomimetics (eserine, pilocarpine),
 - anticholinergic (atropine, hyoscyamine), or
 - **ganglioplegics** (sparteine, nicotine).
- In addition, alkaloids include
 - curare,
 - local anesthetics (cocaine),
 - agents to treat fibrillation (quinidine),
 - antitumor agents (vinblastine, ellipticine),
 - antimalarials (quinine),
 - antibacterials (berberine), and amebicides (emetine).

ALKALOIDS, Uses

These various activities lead to extensive use of alkaloid-containing drugs:

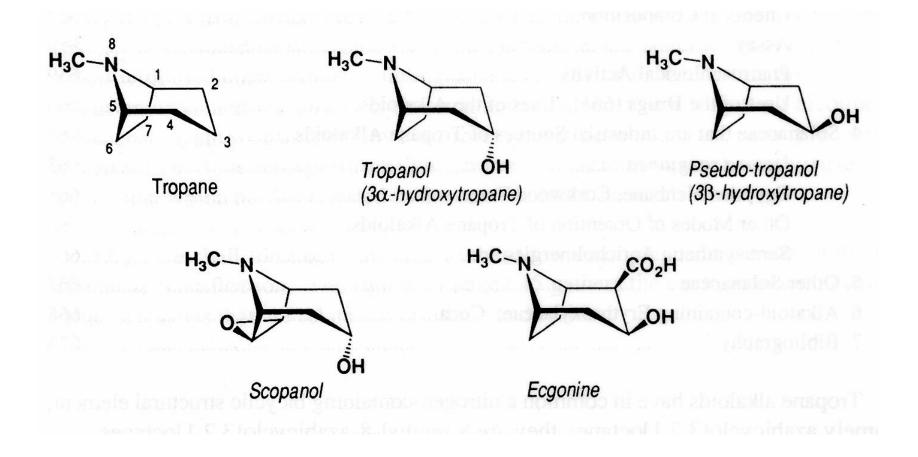
- as galenicals (belladonna, datura, hebane)
- as starting material for industrial extraction: morphine from poppy straw or opium, scopolamine from Duboisia, ajmalicine from Catharanthus roots, vincamine from periwinkle leaves, and quinine from Cinchona bark.
- Some of the **extracted alkaloids may undergo transformations**: codeine is produced mostly by methylating morphine, quinine is converted to quinidine, serpentine to ajmalicine, and tabersonine to vincamine; tropane alkaloids are quaternized.
- In a few rare cases, the industry prefers **direct synthesis**: theophylline and papaverine are easily obtained that way.
- The drive to optimize therapeutic efficacy, has sometimes resulted in achieving **deeper transformations**, or even **total syntheses of analogous molecules**, making use or not of starting materials of natural, plant, or fermentation origin
 - (see especially the derivatives of ergot alkaloids, and those of the binary alkaloids of Catharanthus).

Tropane alkaloids

With a few exceptions, tropane alkaloids are esters of tropane alcohols and of acids of various structures, either aliphatic or aromatic.

A. TROPANOLS

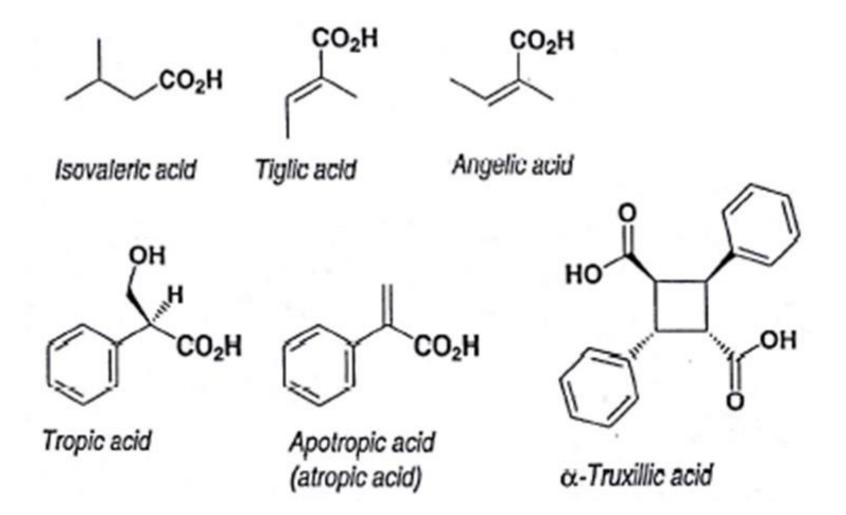
These alcohols fall into two series depending on the orientation of the hydroxyl group at C-3.



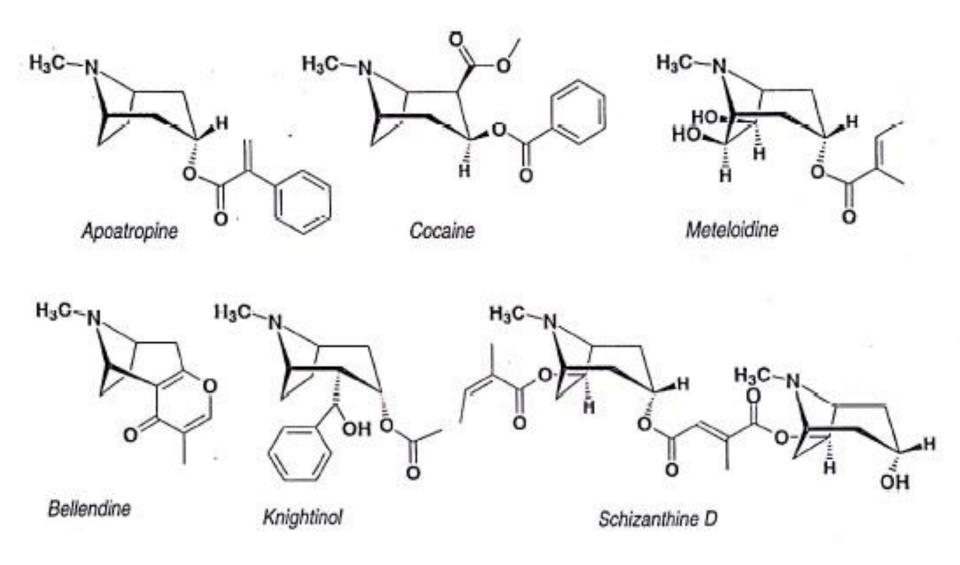
B. ACIDS

The acids may be **aliphatic** (acetic, butiric, icovaleric, 2-methylbutyric, tiglic acid, angelic acid) **or aromatic**.

The aromatic acid may be specific like (S)-(-)-tropic acid, or may be more widely distributed in the plant kingdom like benzoic, phenylacetic, cinnamic acid and their derivatives.

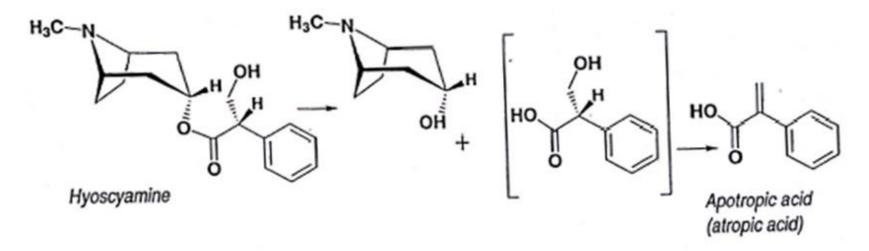


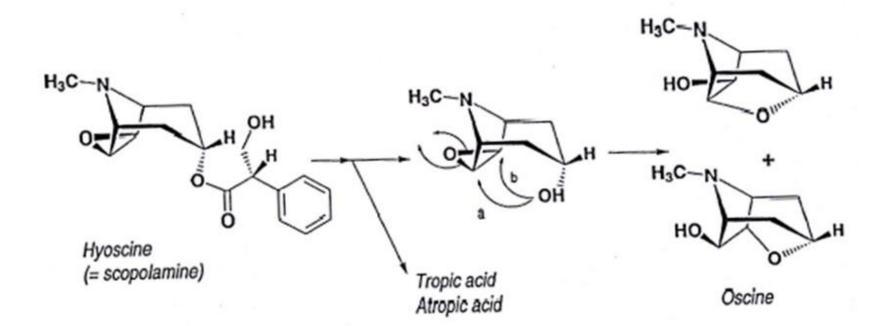
C. ALKALOIDS

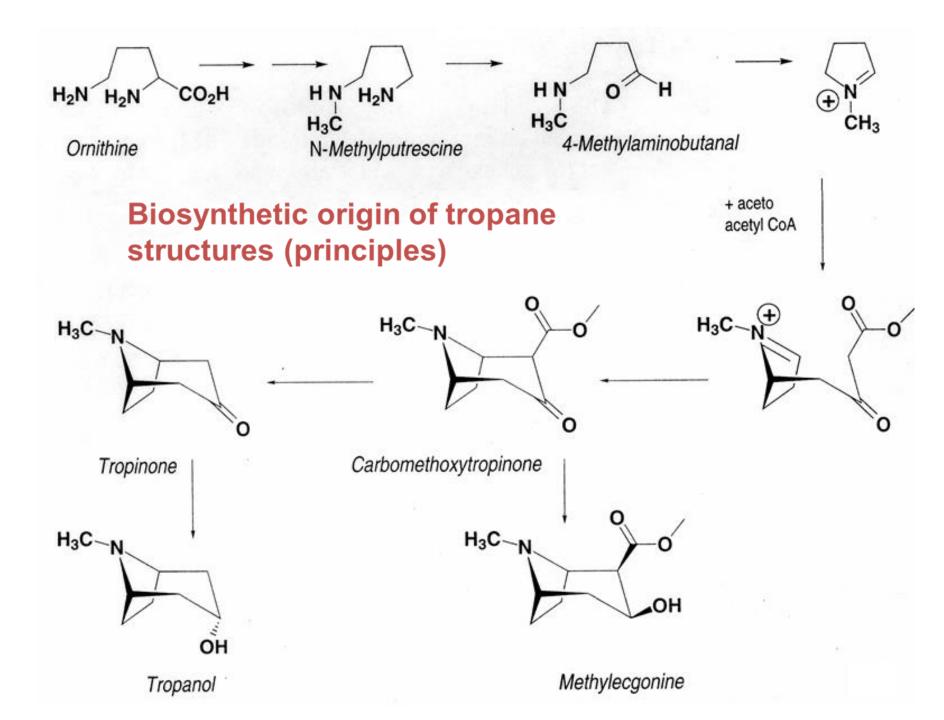


Degradation of tropan ester alkaloids

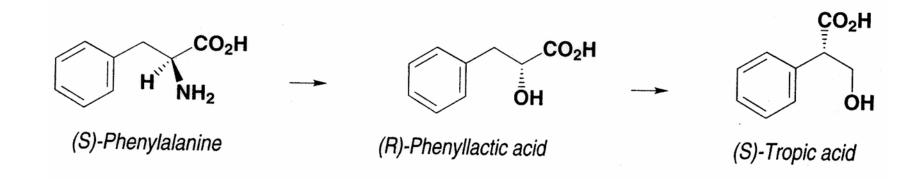
It occures in acidic as well as basic condition







Origin of tropic acid



Deadly Nightshade, *Atropa belladonna* L. (Belladonna), Solanaceae

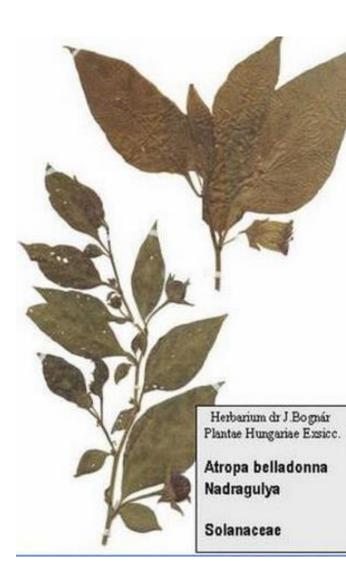






- Indigenous to western Europe. A perennial plant with a rizoma-like root, erect stems (1-1.5m).
- Leaves: alternate on the lower part of the stem , in close pairs near the inflorescence. They are uneven size and not opposite.
- Flowers: normally solitary with a campanulate corolla with purplish-brown or brownish-yellow lobes.
- Fruit: the size of a cherry, shiny black, surraunded at the base by an indeciduous and well-developed calyx.
- (subglobulous bilocular berry)

Belladonnae folium



- Belladonna leaf has an elliptic blade, acuminate at the apex, and attenuate at the base (5-25 x 3-12 cm)
- The pubescence of young leaves only remains near at the veins in the older leaves.
- The secondary veins are at 60° and are anastomosed near the margin.

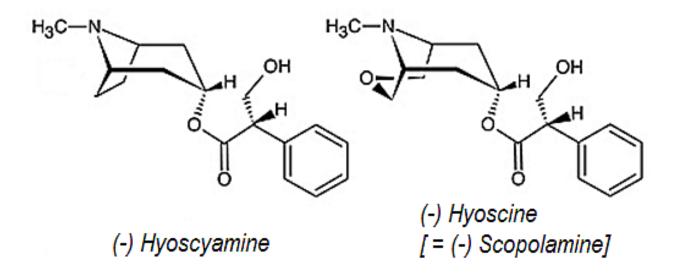
Microscopic characteristic

- Striated cuticle.
- Microsphenoidal crystals in the cells of the parenchymas especially under the palisade layer.
- Trichomes: multicellular, uniseriate, rare covering trichomes.

Belladonnae folium

Chemical composition

- Alkaloid content 0.3-06%.
- **Hyoscyamine as chief constituent** (90%), occures alongside **scopolamine** (2%) and their dehydration products (7%).
- Small quantity of scopoletol (a coumarine).



Belladonnae folium, Uses

Galenicals

- Belladonnae pulvis normatus (Ph. Eur.): titrated to contain 0.28-0.32 % total alkaloids calculated as hyoscyamine.
- Belladonnae leaf dry extract, standardised (Ph. Eur.): 0.95-1.05 % total alkaloids
- Belladonnae leaf tincture, standardised (Ph. Eur.): 0.027-0.033 % total alkaloids

Galenicals in various combinations for the symptomatic treatment of

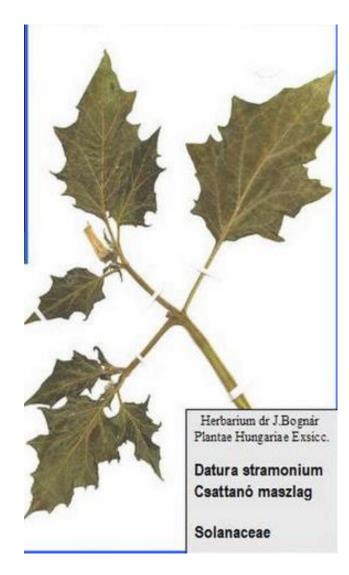
- **unproductive coughs**; sometimes for acute congestion of the throat and larynx.
- as a short term symptomatic treatment for constipation,
- the pain associated with functional problems of the gastrointestinal and biliary tracts.
- In rare antalgic and antineuralgic propietary drugs.
- These combinations have unfavorable benefit-to-risk ratio due to their galenicals contents.

Thorn Apple, *Datura stramonium* L., (Stramonium, Jimson weed), Solanaceae



- It grows abundantly in Europe where it most likes neglected country fields and roadsides.
- A hardly annual species , 0.8-1.2m, has a rounded stem with oval acute leaves deeply divaded in uneven pointy lobes.
- Flowers: solitary, large (8-10 cm long), have a calyx with five sepals pleated longitudinally, a **tubulous** corolla, pleated, and white.
- Fruit: A bilocular capsule with a cover that opens. It covered with tough thorns.

Stramonii folium



- The blade of the stramonium leaf (8-25 x
 7-15 cm) is acumuminate and very often asymmetric at the base.
- The secondary veins are prominent on the lower side and depressed on the upper side.
- The older leaf is practically glabrous, whereas the veins of the young leaves are tomentose. The secondary veins are at 45° and end at the apex of the blade.

Microscopic characteristics

- Epidermal cells with wavy walls and a smooth cuticle.
- Calcium oxalate cluster crystals from 10 to $30\,\mu\text{m}$.
- Numerous conical covering trichomes, often broken and with verrucose wall.

Stramonii folium (Ph. Eur., Ph. Hg VIII.)

Chemical composition

- Minerals 15-18 %
- Total alkaloid content 0.2-05 %.
- At the time of harvest, hyoscyamine and scopolamine represent twothirds and one-thirds of the total alkaloids, respectively.

Uses

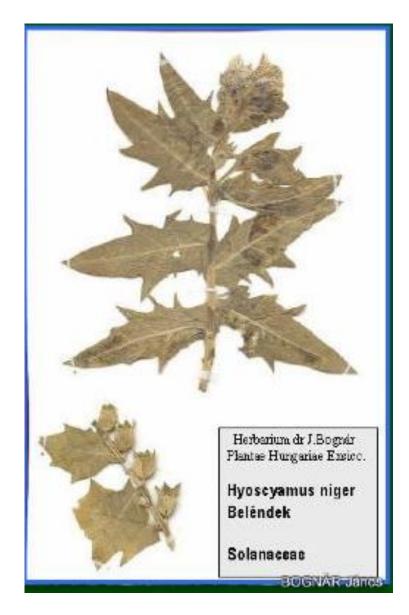
- Stramonii pulvis normatus (Ph. Eur.): official stramonium powder titrated to contain 0.23-0.27 % total alkaloids.
- Preparation of galenicals:
 - only one syrup proposed for the symptomatic treatment of unproductive cough.
 - In the late 1980s, it was still used in cigarettes designed to relive respiratory difficulties.

Henbane, Hyoscyamus niger L.



- Hebane can be annual or biennial depending on the variety.
- Originated from Asia, grows all over Europe and North America.
- The stem is hairy and viscous, either simple (var. annua) or ramnified (var. biennis).
- Leaves are petiolate at the base, sessile or sheating on the stem, with triangular lobes, very hairy, and pale green.
- The flowers grouped into short racemate at the base of a larger bract, have corolla with five lobes incompletly actinomorphic, and grayish-yelollow with purple or purplishblack veins.
 - The pyxidium is surrounded by an indeciduous, enlarged, and hardened calyx with thorny teeth.

Hyoscyami folium



- The henbane leaf can be sessile, in which case it is cordate at the base, or on a short petiole, in which case it is acute.
- The blade (25 x 5-7 cm) is highly pubescent and viscous on both sides, especially near the midribs;
- its margin is irregular and divided in wide triangular lobes.
- The secondary veins form a wide angle with the midrib and run to the apex of the blade.
- Microscopic characteristic: epidermal cells with wavy wall and a smooth cuticule. Calcium oxalate crytals of prisms from 5to 20 um. Trichomes: very numerous glandular trichomes of several types, often fragmented, and numerous covering trichomes.

Hyoscyami folium

Chemical composition

- Minerals (18-20%).
- Total alkaloid content: 0.04-0.15%.
- Hyoscyamine is the chief constituent and the percentage of scopolamine can be high (25 % and more).

Uses

- Hebane is not used much more than stramonium.
- It is an ingredient of combinations, for example with buckthorn, aloe (stimulant laxative), belladonna (gastrointestinal pain), or ephedrine (asthma).

Pharmacological activity of the alkaloids I.

Atropine

- Atropine and hyoscyamine are **parasympatholytics**.
- Hyoscyamine has a stronger activity than the racemic atropine, but it is the latter that is commonly prepared and used.
- Atropin is an inhibitor of the muscarinic receptors of the peripheral organs innervated by the parasympathetic post-ganglionic fibers, and of the central nervous system.
- It acts by competitive and reversible inhibition of acetylcholine binding onto its receptors, and this antagonism leads, in the organs of question, to **sympatomymetic-like effects**.

Autonomic nervous system

- In the **hart** and after temporary bradycardia, atropine **increases the hart rate** by suppressing vagal inhibition.
- The effects on the blood pressure are not marked.
- It **decreases intestinal tone**, the amplitude and frequency of peristaltic contractions, paralyzes the ureters, increases bladder pressure, decreases biliary duct tone, and blocks the bronchoconstricting effect of acetylcholine.
- Saliva, sweat, gastric, pancreatic, bronchial, and lachrymal secretions are decreased.
- It induces a passive mydriasis, a paralysis of the accommodation and an increases in intra-ocular pressure.

Pharmacological activity of the alkaloids II.

Atropine

CNS

- **Toxic doses** cause **substantial excitation**: agitation, disorientation, exaggerated reflexes, hallucinations, delirium, mental confusion, and insomnia;
- at low doses the action is less clear, and tends to be depressant and sedative.

Scopolamine

- The parasympatholytic activity of scopolamine is identical to that of atropine, but much less marked, especially on the myocardium.
- Its effects on the CNS are clear: sedative, depressant, hypnotic with amnesia.
- It potentiates neuroleptics, improves parkinsonism, and is "incapacitating" at high doses.

Brugmansia sanguinea Ruiz et Pav (syn. *Datura sanguinea*), Solanaceae



- **Small tree** characterized by **large flowers** (17-25 cm), with a **tubulous corolla**, yellow and orangy with red veins.
- This plant is **cultivated in Equador** in high altitude (3000 m) areas.
- The **leaves**, which contain about **0.8 % alakloids**, **with scopolamine as chief constituent**, are harvested mechanically three times a year.
- B. sanguinea like other species (Datura innoxia from Mexico, B. suaveolens, B. arborea from Amazonia and Colombia, among others) is traditionally used for its hallucinogenic properties.

Corkwood Tree, Pituri, *Duboisia myoporoides* R.Br., *D. leichardtii* F. Muell. (Solanaceae)



Small trees with

- alternate and narrow leaves ,
- panicles of tubulate white flowers,
- black berries.

Both species are **Australian**: *D. myoporoides* is widespread on the eastern seaboard, whereas *D. leichardiiis* localized around Brisbane.

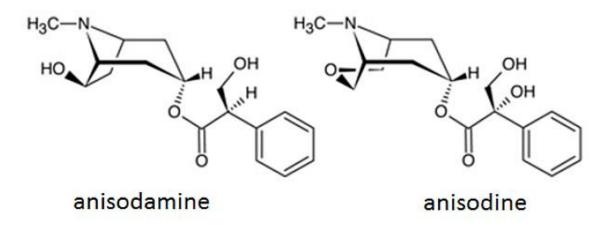
Both species, as well as and their hybrids, are rich in alkaloids (up to 3%) and are cultivated.

Both species are **exploited for the extraction of alkaloids**, which used to be carried out on site for a long time.

At present, the leaves are exported toward Europe, mainly to Germany for extraction.

Anisodus tanguticus (Maxim.) Pasch

- This Chinese plant is an ingredient of traditional anesthetic preparations.
 Its root contains alkaloids:
- Anisodamine, a CNS stimulant, an anticholinergic and antispasmodic.
 - It used **to treat acute enteritis and septic shock** (bacyllary disentery); by dilating the capillaries, it improves microcirculation.
- Anisodine is a CNS depressant, it is antagonized by physostigmine, and chiephly used to treat migraine hedache.



Coca, Cocae folium, Erythroxylum coca, Erythroxylaceae

Coca is a cultivated shrub, pruned to different heights depending on the geographical area (70-80 cm in the Yungas of Bolivia).



Erythroxylum coca Lam. Image processed by Thomas Schoepke www.plant-pictures.de

The branches are reddish , and bear **oval**, entire, and shortly petiolate leaves.

The **flowers** are pentamerous and **yellowish-white**. The **fruit** is a small **red drupe**.

The leaf of the typical species has a slightly acuminate blade (2.5-7.5 x 1.5-4 cm), more or less prominently *marked on the lower side by* two curved lines , which delineate an oval area centered on the midrib.





Coca, Erythoxylum ssp.,

The **Erythroxylum cultivated** to produce leaves **rich in cocaine includes** three taxa, **three varieties linked to two species**:

- *E. coca* Lam var. *coca* grows wild in the Peruvian and Bolivian Andes and cultivated on the damp eastern side of the mountains. The leaves are dark green; the blade is elliptic and wide; its midrib forms a prominent ridge on the upper side.
- *E. novogranatense* (Morris) Hieron var. *novogranatense*. This forest variety grows in Columbia and Venezuela. The leaves are bright yellowish-green, the blade is elliptic and elongated.
- *E. novogranatense* (Morris) Hieron var. *truxillense* (Rusby) Plowman. This variety is characteristic of the dry areas of the north of Peru and of Ecuador. The leaves have an elliptic, very narrow, and pale green blade, and the midrib ridge is flattened.
- The three taxa are thought to represent stages in evolution, which would have *E. coca* var. *coca* as their ancestor; the latter is the only form capable of reproducing without human invention.

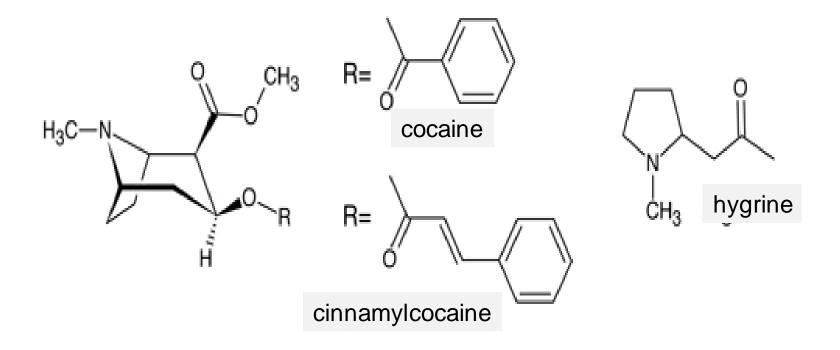


Erythroxylum novogranatense var. *novogranatense*

Cocae folium

Constituents

- Variable quantites of an essential oil including methylsalicilate, flavonoids, tannins.
- Alkaloids (0.5-1.5 %): Cocaine (=methylbenzoylecgonine, 30-50%), cinnamylcocaine (=methylcinnamylecgonine), truxillines (esters of a dicinnamic acid), several pyrrolidines (hygrine, cuscohygrine).



Cocae folium, Pharmacological Properties

Cocaine

- Local (contact) anesthetic. It blocks ion channels in neuronal membranes, and interrupts the propagation of action potential corresponding to the sensory message.
- Sympathomimetic. It acts as an adrenergic stimulant by blocking the reuptake of dopamine, and noradrenaline at the presynaptic neuron by binding to their transporters.
- This adrenergic stimulation causes hyperthermia , mydriasis, and vasoconstriction of most of the blood vessels, which increases resistance and contributes to increasing blood pressure.
- **Centrally**, the stimulation **results in a sensation of euphoria** with intellectual stimulation, decreased inhibition, hyperactivity, and of effects sought by drug addicts.
- The depletion which follows the reuptake blockade explains the **short term depressant effect** (psychic and physical asthenia, respiratory and vasomotor depression) and **rapid development of an intense psychic dependence** which is reinforced by further abuse.
- Cocaine does not induce physical dependence.

Cocae folium

Uses

Neither coca leaf nor its galenicals are used any more, but **the leaves** are still used **to extract cocaine.**

In the United States, **cocaine** is used **in combinations** (phenol, menthol, cocaine) **for local anaeshesia**, for example to stich small wounds.

Traditional uses

- As a masticatory, a very ancient habit; the coca leaf is chewed, and added alkalis facilitate the release of cocaine.
- In infusion; the common form is a **tea bag** which yields a strikingly aromatic infusion, consumed like coffee or tea (*mate de coca*).

Illicit Use of Cocaine

- **Cocaine hydrochloride** is generally "snorted" by the intranasal route, and less often used by IV injection.
- Cocaine intake causes **euphoria**, intellectual stimulation, hyperactivity, a feeling of hyperlucidity, and acceleration in elaboration of ideas.
- Its activity resembles that of amphetamines, and also manifests itself by a **decrease in fatigue, insomnia, anorexia, and increased talkativeness**,
- but also by irritability, altered sensations and impaired judgement, physical exhaustion, and emotional depression.
- Cocaine use commonly causes severe headaches and sometimes causes convulsions;
- **delusions** and **hallucinations** suggesting a serious paranoid psychosis are also described.
- Another effect is compulsive scratching, and *difficulties with verbal expression and memorization* are common.
- The most serious complications are cardiovascular: hypertensive emergency, myocardial ischemia degenerating into an infarct, cerebral hemorrhage.
- Massive overdose is characterized by coma, convulsions, and cardiac alterations. The risk is higher in alcohol users: the liver esterases transesterify cocaine into cocaethylene, which is particularly toxic.

Illicit Use of Cocaine

- **Coca paste,** the initial product of the extraction of the leaves, **contains from 40 to 70 % cocaine** (extraction of the leaves with sulfuric acid, alkalinization with carbonate, dissolution of the free base in kerosene). The past is smoked. Kerosene and other residual solvents impart their own toxicity to the preparation.
- Cocaine is also smoked (this is "freebasing"). The smoked forms (pure base) have intense effects with a rapid onset, but these effects do not last; the profound depression which follows drives the user to take the drug again, and dependence sets in very rapidly.

Thank you for your attention !