

Stephen J. Chapman* and Arabo A. Avanes

*Correspondence to: Isomer Design, 4103-210 Victoria St, Toronto, ON, M5B 2R3, Canada. E-mail: SJChapman@isomerdesign.com

¹H NMR spectra of 28 alleged psychedelic phenylethanamines from 15 grey-market internet vendors across North America and Europe were acquired and compared. Members from each of the principal phenylethanamine families were analyzed: eleven *para*-substituted 2,5-dimethoxyphenylethanamines (the 2C and 2C-T series); four *para*-substituted 3,5-dimethoxyphenylethanamines (mescaline analogues); two β -substituted phenylethanamines; and ten *N*-substituted phenylethanamines with a 2-methoxybenzyl (NBOMe), 2-hydroxybenzyl (NBOH), or 2,3-methylenedioxybenzyl (NBMD) amine moiety. ¹H NMR spectra for some of these compounds have not been previously reported to our knowledge. Others have reported on the composition of “mystery pills,” single-dose formulations obtained from retail shops and websites. We believe this is the first published survey of bulk “research chemicals” marketed and sold as such. Only one analyte was unequivocally misrepresented. This collection of experimentally uniform spectra may help forensic and harm-reduction organizations identify these compounds, some of which appear only sporadically. The complete spectra are provided as supplementary data.^[1]

Keywords: ¹H NMR, drug checking, grey markets, research chemicals, phenylethanamines, *N*-benzyl phenylethanamines, PiHKAL
DOI: <http://dx.doi.org/10.16889/isomerdesign-1> Published: 1 August 2015 Version: 1.03

*“Once you get a serious spectrum collection,
the tendency is to push it as far as you can.”¹*

Introduction

Psychedelic phenylethanamines² have been known and studied for decades, generating much interest and enthusiasm in some quarters — and corresponding alarm in others.^[3–6] Though drawing less scrutiny than better known psychedelics like LSD or psilocybin, strictly speaking these are not novel materials.^[7,8] In contrast, *N*-benzyl phenylethanamines progressed from earliest published reports^[9–14] and subsequent research^[15,16] to ready grey-market availability and ultimately prohibition — in the UK^[17,18] and the USA^[19] — in less than 20 years.

Here we report the ¹H NMR spectrum of 28 grey-market phenyl ethanamines and *N*-benzyl phenylethanamines. The spectra of ca. 30 grey-market tryptamines will be the subject of a later report.

The advent and rapid expansion of electronic commerce in grey-market research chemicals has been startling.^[20–22] Hundreds of formerly obscure substances, once unobtainable to most in any practical sense are now within reach.^[23] For some, legal and affordable reach as well. A wide-ranging review of the evolving new-drug landscape is provided by Brandt et al.^[24]

Nevertheless, an inherent weakness of grey markets is the absence of regulatory oversight. Fleeting, unreliable, or fraudulent vendors are not uncommon. The reputation of a well-regarded source may, gradually or abruptly, decline. The slide from principled vendor to scam artist is well worn. For the customer, authenticating grey-market products is essential — and a considerable challenge.

Few organizations exist to meet this need. EcstasyData.org, a leading drug-analysis service, has traditionally focused on testing single-dose formulations (e.g. a pill or capsule) held out as “ecstasy.” More recently, they expanded their mandate to include research chemicals.³ However, at \$100 per analyte the cost seemed prohibitive for the summary level of analysis provided — a restriction imposed by the Drug Enforcement Administration (DEA).⁴

Fortunately, the University of Toronto offers several forms of analysis for external clients including NMR and mass spectrometry (MS). We wondered if “walk-in” ¹H NMR analysis combined with contemporary spectral-analysis software might allow even inexperienced investigators to verify the composition of grey-market research chemicals with a reasonable degree of confidence. The attraction of ¹H NMR over MS is threefold: (1) the cost is significantly lower; less than 25 percent, (2) it provides sufficient information to potentially

¹ *pace* Hunter S. Thompson.

² Systematic names follow IUPAC Recommendations and Preferred Names 2013 except where noted. Sadly, the beloved and familiar contraction *phenethyl* has officially fallen from grace (i.e. deprecated).^[2]

³ Isomer Design is a supporting partner in EcstasyData.org.

⁴ Imposed by “an unpublished administrative rule.” See EcstasyData.org.



identify an unknown in the absence of a reference spectrum,^[25] and (3) aryl regioisomers are more readily differentiated.

Simply, we wanted to know if the grey-market research chemicals being sold were in fact “as advertised.” Would the spectrum be consistent with the substance as alleged? Was this powder labelled *x*, actually *x*, or not? In November 2012, after an engaging discussion with Timothy Burrow, Director of the University of Toronto NMR Facility, we began our investigation to confirm the identity of grey-market research chemicals.

Limitations and bias

Our objective was chiefly to *check* the identity of our unknowns, not unlike how presumptive spot-colour tests are used.^[26] It was sufficient to assign a pass/fail grade to an unknown; spectra consistent with published results passed.¹

We have included every phenylethamine spectrum acquired — nothing has been excluded. Our analytes were purchased over several years from 15 vendors across North America and Europe. Though we consider this to be a fair representation of the grey market during that time it is not without potential selection bias. While no official regulatory oversight exists for this market, anecdotal reports posted at SafeOrScam.com² provided some guidance. Patently disreputable vendors were excluded. Our analytes were not selected *at random* and our findings should be interpreted accordingly.

The “purity” values we report were provided by the verification module of the MestReNova software. Though we do not suggest these values necessarily reflect actual analyte purity, they are comparable in magnitude and implied precision (misleading or not) to vendor-supplied values.³

A report by Hays and Cassale^[28] became available to us only recently. We regret we were unable to include their extensive findings among the cited literature.

Experimental

Grey-market research chemicals were purchased from internet vendors in Europe, the United States, and Canada, 2009–2014.

Commercially available 2-phenylethan-1-amine HCl (PEA) was purchased from AK Scientific, Inc., Union City, CA. Two *N*-benzylidene phenylethanamines were purchased from Cayman Chemical, Ann Arbor, MI. Licensed software from

Mestrelab Research SL, Santiago de Compostela, Spain was used for spectral analysis. Preferred IUPAC names were constructed using licenced software from ACD/Labs, Toronto, Canada.

Analytical data was supplied by Timothy Burrow from November 2012 through August 2014. ¹H NMR spectra for analytes **P0** and **P17** were recorded on an Agilent DD2 NMR spectrometer operating at 699.8 MHz and equipped with an HFCN Cold Probe. All other analytes were recorded on an Agilent DD2 NMR spectrometer operating at 499.67 MHz and equipped with an XSens Cold Probe.

Approximately 5 mg of each analyte was placed in a capped glass vial, labelled, and submitted for analysis. Except where noted spectra were acquired in deuterated dimethyl sulfoxide (DMSO-*d*₆) from single-use ampoules to minimize the presence of water. DMSO-*d*₆ was chosen to dissolve analytes in free base or salt form along with any impurities which might be present. Dissolution was reportedly facile and complete in every case, leaving no residual insoluble material. Dissolved analytes were loaded without filtering into a 3 mm NMR sample tubes. Spectra acquired in DMSO-*d*₆ were referenced to the solvent signal at $\delta = 2.50$ ppm.

Results

Analyte structures and codes are shown in Figures 1–3. Analyte codes, substitution patterns, and preferred IUPAC names (PIN) are listed in Table 1. ¹H NMR spectral data is summarized in Table 2. Multiplets for moieties common to all scaffolds are grouped together in Table 3. Multiplets for moieties common only to the *N*-benzyl and *N*-benzylidene PEA scaffolds are grouped together in Table 4, ordered by shift. Multiplets for *para*-substituents, β -substituents, and atypical moieties are detailed in Table 5.

¹ If no published spectrum could be found we used the Force of MestReNova software to predict one. *How* to use the Force without succumbing to the Dark Side was revealed in an e-mail from Jedi Master Santi Domínguez, 2013.

² SafeOrScam.com has closed, reportedly sacrificed following an attack ca. 8 July 2015. In its place, ostensibly, ScamLogs.com has since appeared.^[27]

³ The *alleged* purity of our analytes routinely implies a precision of $\pm 0.01\%$.

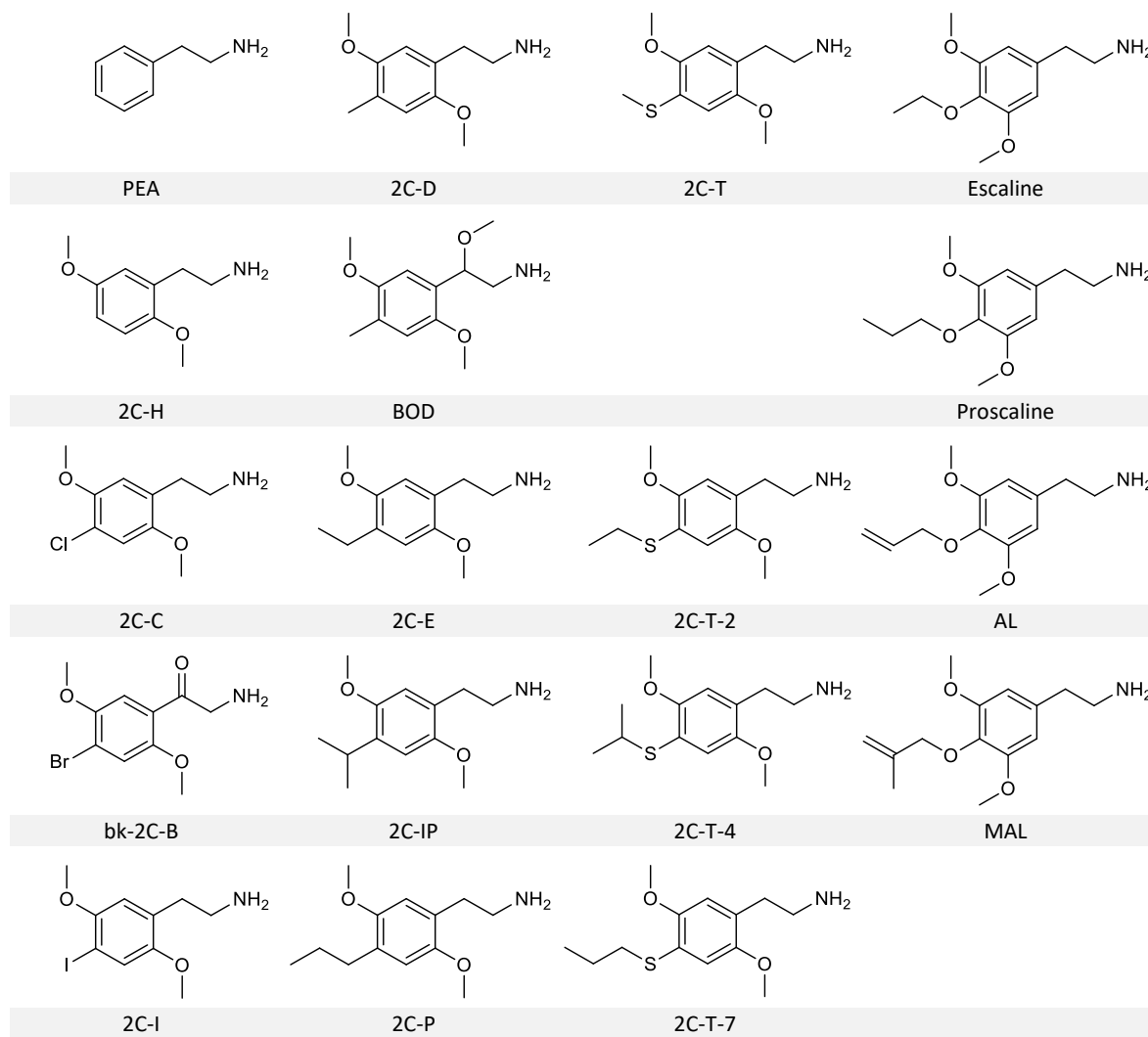


Figure 1: Structures of *N*-unsubstituted (*P* series) analytes

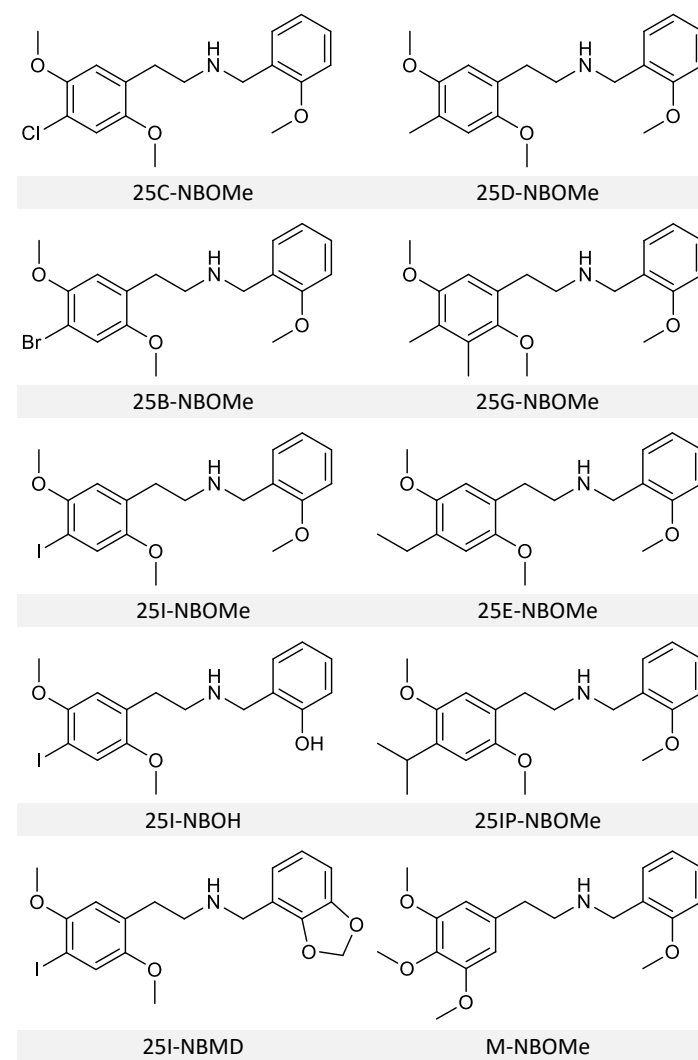


Figure 2: Structures of *N*-benzyl (*B* series) analytes

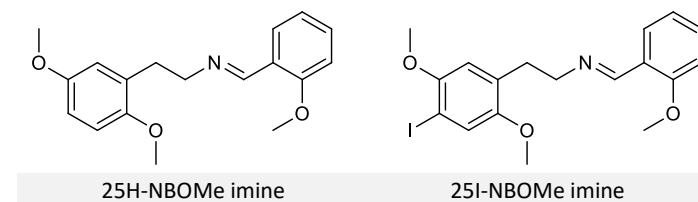
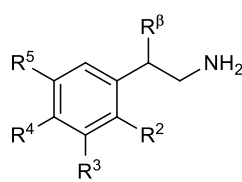
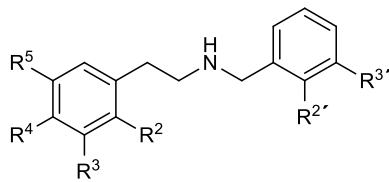
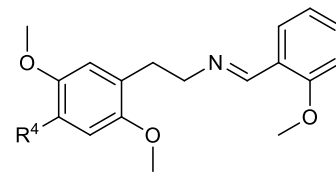


Figure 3: Structures of *N*-benzylidene (*IM* series) analytes

Table 1: Names and scaffold substitution patterns of analytes

**P****B****IM****Scaffold P**

Code	R ²	R ³	R ⁴	R ⁵	R ^β	Preferred IUPAC name
PEA	H	H	H	H	H	2-phenylethan-1-amine
2C-H	OCH ₃	H	H	OCH ₃	H	2-(2,5-dimethoxyphenyl)ethan-1-amine
2C-C	OCH ₃	H	Cl	OCH ₃	H	2-(4-chloro-2,5-dimethoxyphenyl)ethan-1-amine
bk-2C-B	OCH ₃	H	Br	OCH ₃	=O	2-amino-1-(4-bromo-2,5-dimethoxyphenyl)ethan-1-one
2C-I	OCH ₃	H	I	OCH ₃	H	2-(4-iodo-2,5-dimethoxyphenyl)ethan-1-amine
2C-D	OCH ₃	H	CH ₃	OCH ₃	H	2-(2,5-dimethoxy-4-methylphenyl)ethan-1-amine
BOD	OCH ₃	H	CH ₃	OCH ₃	OCH ₃	2-(2,5-dimethoxy-4-methylphenyl)-2-methoxyethan-1-amine
2C-E	OCH ₃	H	CH ₂ CH ₃	OCH ₃	H	2-(4-ethyl-2,5-dimethoxyphenyl)ethan-1-amine
2C-IP	OCH ₃	H	CH(CH ₃) ₂	OCH ₃	H	2-[2,5-dimethoxy-4-(propan-2-yl)phenyl]ethan-1-amine
2C-P	OCH ₃	H	CH ₂ CH ₂ CH ₃	OCH ₃	H	2-(2,5-dimethoxy-4-propylphenyl)ethan-1-amine
2C-T	OCH ₃	H	SCH ₃	OCH ₃	H	2-[2,5-dimethoxy-4-(methylsulfanyl)phenyl]ethan-1-amine
2C-T-2	OCH ₃	H	SCH ₂ CH ₃	OCH ₃	H	2-[4-(ethylsulfanyl)-2,5-dimethoxyphenyl]ethan-1-amine
2C-T-4	OCH ₃	H	SCH(CH ₃) ₂	OCH ₃	H	2-{2,5-dimethoxy-4-[(propan-2-yl)sulfanyl]phenyl}ethan-1-amine
2C-T-7	OCH ₃	H	SCH ₂ CH ₂ CH ₃	OCH ₃	H	2-[2,5-dimethoxy-4-(propylsulfanyl)phenyl]ethan-1-amine
Escaline	H	OCH ₃	OCH ₂ CH ₃	OCH ₃	H	2-(4-ethoxy-3,5-dimethoxyphenyl)ethan-1-amine
Proscaline	H	OCH ₃	OCH ₂ CH ₂ CH ₃	OCH ₃	H	2-(3,5-dimethoxy-4-propoxyphenyl)ethan-1-amine
AL	H	OCH ₃	OCH ₂ CHCH ₂	OCH ₃	H	2-{3,5-dimethoxy-4-[(prop-2-en-1-yl)oxy]phenyl}ethan-1-amine
MAL	H	OCH ₃	OCH ₂ C(CH ₃)CH ₂	OCH ₃	H	2-{3,5-dimethoxy-4-[(2-methylprop-2-en-1-yl)oxy]phenyl}ethan-1-amine

Scaffold B

Code	R ²	R ³	R ⁴	R ⁵	R ^{2'}	R ^{3'}	Preferred IUPAC name
25C-NBOMe	OCH ₃	H	Cl	OCH ₃	OCH ₃	H	2-(4-chloro-2,5-dimethoxyphenyl)-N-[(2-methoxyphenyl)methyl]ethan-1-amine
25B-NBOMe	OCH ₃	H	Br	OCH ₃	OCH ₃	H	2-(4-bromo-2,5-dimethoxyphenyl)-N-[(2-methoxyphenyl)methyl]ethan-1-amine
25I-NBOMe	OCH ₃	H	I	OCH ₃	OCH ₃	H	2-(4-iodo-2,5-dimethoxyphenyl)-N-[(2-methoxyphenyl)methyl]ethan-1-amine
25I-NBOH	OCH ₃	H	I	OCH ₃	OH	H	2-({[2-(4-iodo-2,5-dimethoxyphenyl)ethyl]amino}methyl)phenol
25I-NBMD	OCH ₃	H	I	OCH ₃	-O-CH ₂ -O-	H	N-[(2 <i>H</i> -1,3-benzodioxol-4-yl)methyl]-2-(4-iodo-2,5-dimethoxyphenyl)ethan-1-amine
25D-NBOMe	OCH ₃	H	CH ₃	OCH ₃	OCH ₃	H	2-(2,5-dimethoxy-4-methylphenyl)-N-[(2-methoxyphenyl)methyl]ethan-1-amine
25G-NBOMe	OCH ₃	CH ₃	CH ₃	OCH ₃	OCH ₃	H	2-(2,5-dimethoxy-3,4-dimethylphenyl)-N-[(2-methoxyphenyl)methyl]ethan-1-amine
25E-NBOMe	OCH ₃	H	CH ₂ CH ₃	OCH ₃	OCH ₃	H	2-(4-ethyl-2,5-dimethoxyphenyl)-N-[(2-methoxyphenyl)methyl]ethan-1-amine
25IP-NBOMe	OCH ₃	H	CH(CH ₃) ₂	OCH ₃	OCH ₃	H	2-[2,5-dimethoxy-4-(propan-2-yl)phenyl]-N-[(2-methoxyphenyl)methyl]ethan-1-amine
M-NBOMe	H	OCH ₃	OCH ₃	OCH ₃	OCH ₃	H	N-[(2-methoxyphenyl)methyl]-2-(3,4,5-trimethoxyphenyl)ethan-1-amine

Scaffold IM

Code	R ⁴	Preferred IUPAC name
25H-NBOMe imine	H	N-[2-(2,5-dimethoxyphenyl)ethyl]-1-(2-methoxyphenyl)methanimine
25I-NBOMe imine	I	N-[2-(4-iodo-2,5-dimethoxyphenyl)ethyl]-1-(2-methoxyphenyl)methanimine

Table 2: ¹H NMR chemical shift, labelled formulation, and calculated purity of analytes

Analyte ^a	Form ^b	Purity ^c	¹ H NMR ^d	
P0	PEA · HCl	HCl	98 ^e	700 MHz DMSO- <i>d</i> ₆ δ 8.29 (s, 3H), 7.34 – 7.21 (m, 5H), 3.02 – 2.97 (m, 2H), 2.95 – 2.89 (m, 2H)
P1	2C-H base	U	98.16	500 MHz DMSO- <i>d</i> ₆ δ 6.88 – 6.81 (m, 1H), 6.76 – 6.68 (m, 2H), 3.71 (s, 3H), 3.68 (s, 3H), 2.73 – 2.66 (m, 2H), 2.63 – 2.55 (m, 2H), 1.32 (vb s, 2H)
P1^f	2C-H · HCl	–	98.46	500 MHz DMSO- <i>d</i> ₆ δ 8.19 (br s, 3H), 6.94 – 6.87 (m, 1H), 6.82 – 6.76 (m, 2H), 3.74 (s, 3H), 3.69 (s, 3H), 2.98 – 2.91 (m, 2H), 2.88 – 2.81 (m, 2H)
P2	2C-D	U	96.44	500 MHz DMSO- <i>d</i> ₆ δ 8.15 (br s, 3H), 6.82 (d, <i>J</i> = 0.8 Hz, 1H), 6.78 (s, 1H), 3.73 (s, 6H), 2.97 – 2.89 (m, 2H), 2.87 – 2.80 (m, 2H), 2.13 (d, <i>J</i> = 0.7 Hz, 3H)
P3	2C-E	HCl	97.50	500 MHz DMSO- <i>d</i> ₆ δ 8.02 (br s, 3H), 6.81 (s, 1H), 6.79 (s, 1H), 3.74 (s, 3H), 3.74 (s, 3H), 3.00 – 2.91 (m, 2H), 2.86 – 2.79 (m, 2H), 2.54 (q, <i>J</i> = 7.5 Hz, 2H), 1.11 (t, <i>J</i> = 7.5 Hz, 3H)
P4	2C-P	U	93.05	500 MHz DMSO- <i>d</i> ₆ δ 8.04 (br s, 3H), 6.79 (s, 2H), 3.74 (s, 3H), 3.73 (s, 3H), 2.99 – 2.90 (m, 2H), 2.86 – 2.79 (m, 2H), 2.51 – 2.45 (m, 2H), 1.60 – 1.45 (m, 2H), 0.89 (t, <i>J</i> = 7.4 Hz, 3H)
P4^g	2C-P	U	–	500 MHz DMF- <i>d</i> ₇ δ 8.64 (br s, 3H), 6.99 (s, 1H), 6.87 (s, 1H), 3.82 (s, 3H), 3.80 (s, 3H), 3.26 – 3.19 (m, 2H), 3.10 – 3.03 (m, 2H), 2.59 – 2.52 (m, 2H), 1.63 – 1.52 (m, 2H), 0.92 (t, <i>J</i> = 7.4 Hz, 3H)
P5	2C-IP	HCl	98.12	500 MHz DMSO- <i>d</i> ₆ δ 8.00 (br s, 3H), 6.81 (s, 1H), 6.79 (s, 1H), 3.76 (s, 3H), 3.74 (s, 3H), 3.21 (hept, <i>J</i> = 6.9 Hz, 1H), 2.99 – 2.92 (m, 2H), 2.85 – 2.78 (m, 2H), 1.15 (d, <i>J</i> = 6.9 Hz, 6H)
P6a	2C-E ^h	FB	96.91	500 MHz DMSO- <i>d</i> ₆ δ 8.03 (br s, 3H), 6.81 (s, 1H), 6.79 (s, 1H), 3.74 (s, 3H), 3.74 (s, 3H), 2.98 – 2.91 (m, 2H), 2.86 – 2.79 (m, 2H), 2.54 (q, <i>J</i> = 7.5 Hz, 2H), 1.11 (t, <i>J</i> = 7.5 Hz, 3H)
P6b	2C-C	C	94.94	500 MHz DMSO- <i>d</i> ₆ δ 8.18 (br s, 3H), 7.07 (s, 1H), 7.05 (s, 1H), 3.80 (s, 3H), 3.76 (s, 3H), 3.00 – 2.93 (m, 2H), 2.90 – 2.83 (m, 2H)
P7	2C-I	HCl	97.36	500 MHz DMSO- <i>d</i> ₆ δ 7.99 (br s, 3H), 7.33 (s, 1H), 6.90 (s, 1H), 3.77 (s, 3H), 3.76 (s, 3H), 3.00 – 2.93 (m, 2H), 2.87 – 2.80 (m, 2H)
P8	2C-T	FB	96.61	500 MHz DMSO- <i>d</i> ₆ δ 7.97 (br s, 3H), 6.83 (s, 1H), 6.76 (s, 1H), 3.80 (s, 3H), 3.76 (s, 3H), 2.98 – 2.91 (m, 2H), 2.86 – 2.79 (m, 2H), 2.41 (s, 3H)
P9	2C-T-2	HCl	96.38	500 MHz DMSO- <i>d</i> ₆ δ 7.98 (br s, 3H), 6.85 (s, 1H), 6.83 (s, 1H), 3.78 (s, 3H), 3.76 (s, 3H), 3.01 – 2.92 (m, 2H), 2.92 (q, <i>J</i> = 7.3 Hz, 2H), 2.86 – 2.79 (m, 2H), 1.23 (t, <i>J</i> = 7.3 Hz, 3H)
P10	2C-T-4	U	97.02	500 MHz DMSO- <i>d</i> ₆ δ 7.98 (br s, 3H), 6.91 (s, 1H), 6.87 (s, 1H), 3.77 (s, 3H), 3.76 (s, 3H), 3.54 (hept, <i>J</i> = 6.6 Hz, 1H), 3.01 – 2.93 (m, 2H), 2.87 – 2.80 (m, 2H), 1.21 (d, <i>J</i> = 6.6 Hz, 6H)
P11	2C-T-7	FB	97.81	500 MHz DMSO- <i>d</i> ₆ δ 8.02 (br s, 3H), 6.85 (s, 1H), 6.82 (s, 1H), 3.77 (s, 3H), 3.76 (s, 3H), 2.99 – 2.92 (m, 2H), 2.88 (t, <i>J</i> = 7.1 Hz, 2H), 2.86 – 2.79 (m, 2H), 1.58 (h, <i>J</i> = 7.3 Hz, 2H), 0.98 (t, <i>J</i> = 7.3 Hz, 3H)
P12	Escaline	FB	96.92	500 MHz DMSO- <i>d</i> ₆ δ 8.22 (br s, 3H), 6.56 (s, 2H), 3.85 (q, <i>J</i> = 7.0 Hz, 2H), 3.75 (s, 6H), 3.06 – 2.99 (m, 2H), 2.89 – 2.81 (m, 2H), 1.20 (t, <i>J</i> = 7.0 Hz, 3H)
P13a	Proscaline	N/A	96.55	500 MHz DMSO- <i>d</i> ₆ δ 8.05 (br s, 3H), 6.56 (s, 2H), 3.79 – 3.72 (m, 8H), 3.07 – 3.00 (m, 2H), 2.85 – 2.78 (m, 2H), 1.61 (h, <i>J</i> = 7.1 Hz, 2H), 0.94 (t, <i>J</i> = 7.4 Hz, 3H)
P13b	Proscaline	FB	96.72	500 MHz DMSO- <i>d</i> ₆ δ 8.20 (br s, 3H), 6.56 (s, 2H), 3.80 – 3.71 (m, 8H), 3.06 – 2.98 (m, 2H), 2.89 – 2.80 (m, 2H), 1.66 – 1.55 (m, 2H), 0.93 (t, <i>J</i> = 7.4 Hz, 3H)
P14	AL	C	95.26	500 MHz DMSO- <i>d</i> ₆ δ 8.13 (br s, 3H), 6.57 (s, 2H), 5.98 (ddt, <i>J</i> = 17.3, 10.4, 5.6 Hz, 1H), 5.28 (dq, <i>J</i> = 17.3, 1.8 Hz, 1H), 5.17 – 5.10 (m, 1H), 4.35 (dt, <i>J</i> = 5.6, 1.6 Hz, 2H), 3.76 (s, 6H), 3.06 – 2.99 (m, 2H), 2.87 – 2.80 (m, 2H)
P15	MAL	C	92.08	500 MHz DMSO- <i>d</i> ₆ δ 8.03 (br s, 3H), 6.56 (s, 2H), 5.02 – 4.97 (m, 1H), 4.89 – 4.83 (m, 1H), 4.27 – 4.22 (m, 2H), 3.76 (s, 6H), 3.06 – 3.00 (m, 2H), 2.86 – 2.78 (m, 2H), 1.81 – 1.76 (m, 3H)
P16	BOD	HCl	97.02	500 MHz DMSO- <i>d</i> ₆ δ 8.14 (s, 3H), 6.91 (d, <i>J</i> = 0.9 Hz, 1H), 6.82 (s, 1H), 4.78 (dd, <i>J</i> = 8.9, 3.9 Hz, 1H), 3.75 (s, 3H), 3.75 (s, 3H), 3.21 (s, 3H), 2.96 – 2.82 (m, 2H), 2.16 (d, <i>J</i> = 0.9 Hz, 3H)
P17	bk-2C-B	HCl	89.78	700 MHz DMSO- <i>d</i> ₆ δ 8.28 (br s, 3H), 7.59 (s, 1H), 7.43 (s, 1H), 4.31 (q, <i>J</i> = 5.2 Hz, 2H), 3.94 (s, 3H), 3.86 (s, 3H)

Table 2: ¹H NMR chemical shift, labelled formulation, and calculated purity of analytes

Analyte ^a	Form ^b	Purity ^c	¹ H NMR ^d
B1	25B-NBOMe HCl	98.32	500 MHz DMSO- <i>d</i> ₆ δ 9.18 (br s, 2H), 7.49 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H), 7.19 (s, 1H), 7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H), 7.02 (s, 1H), 7.00 (td, <i>J</i> = 7.4, 1.1 Hz, 1H), 4.11 (s, 2H), 3.83 (s, 3H), 3.79 (s, 3H), 3.75 (s, 3H), 3.10 – 3.03 (m, 2H), 3.00 – 2.93 (m, 2H)
B2	25C-NBOMe HCl	98.56	500 MHz DMSO- <i>d</i> ₆ δ 9.23 (br s, 2H), 7.50 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.2, 7.4, 1.7 Hz, 1H), 7.14 – 7.05 (m, 1H), 7.08 (s, 1H), 7.04 (s, 1H), 7.00 (td, <i>J</i> = 7.5, 1.1 Hz, 1H), 4.11 (s, 2H), 3.83 (s, 3H), 3.80 (s, 3H), 3.74 (s, 3H), 3.10 – 3.01 (m, 2H), 3.03 – 2.94 (m, 2H)
B3	25I-NBOMe HCl	97.59	500 MHz DMSO- <i>d</i> ₆ δ 9.15 (br s, 2H), 7.48 (dd, <i>J</i> = 7.5, 1.8 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H), 7.32 (s, 1H), 7.08 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H), 6.99 (td, <i>J</i> = 7.4, 1.1 Hz, 1H), 6.90 (s, 1H), 4.11 (s, 2H), 3.83 (s, 3H), 3.77 (s, 3H), 3.73 (s, 2H), 3.09 – 3.00 (m, 2H), 2.99 – 2.92 (m, 2H)
B4	25I-NBOH HCl	96.72	500 MHz DMSO- <i>d</i> ₆ δ 9.22 (vb s, 3H), 7.39 (dd, <i>J</i> = 7.6, 1.7 Hz, 1H), 7.32 (s, 1H), 7.23 (ddd, <i>J</i> = 8.1, 7.3, 1.7 Hz, 1H), 6.97 (dd, <i>J</i> = 8.2, 1.1 Hz, 1H), 6.89 (s, 1H), 6.84 (td, <i>J</i> = 7.4, 1.1 Hz, 1H), 4.09 (s, 2H), 3.77 (s, 3H), 3.74 (s, 3H), 3.09 – 3.02 (m, 2H), 3.00 – 2.91 (m, 2H)
B5ⁱ	25I-NBMD FB	96.42	500 MHz DMSO- <i>d</i> ₆ δ 7.25 (s, 1H), 6.84 (s, 1H), 6.86 – 6.78 (m, 1H), 6.81 – 6.72 (m, 2H), 5.96 (s, 2H), 3.73 (s, 3H), 3.71 (d, <i>J</i> = 0.3 Hz, 3H), 3.66 (d, <i>J</i> = 0.6 Hz, 2H), 2.69 – 2.65 (m, 4H), 1.98 (br s, 1H)
B6	25D-NBOMe HCl	97.39	500 MHz DMSO- <i>d</i> ₆ δ 9.18 (br s, 2H), 7.49 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H), 7.08 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H), 7.00 (td, <i>J</i> = 7.5, 1.1 Hz, 1H), 6.88 – 6.79 (m, 1H), 6.77 (s, 1H), 4.11 (s, 2H), 3.83 (s, 3H), 3.73 (s, 3H), 3.71 (s, 3H), 3.10 – 2.96 (m, 2H), 3.00 – 2.88 (m, 2H), 2.13 (d, <i>J</i> = 0.7 Hz, 3H)
B7	25E-NBOMe HCl	98.67	500 MHz DMSO- <i>d</i> ₆ δ 9.20 (br s, 2H), 7.49 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H), 7.09 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H), 7.00 (td, <i>J</i> = 7.5, 1.0 Hz, 1H), 6.80 (s, 1H), 6.78 (s, 1H), 4.12 (s, 2H), 3.83 (s, 3H), 3.73 (s, 3H), 3.72 (s, 3H), 3.07 – 2.99 (m, 2H), 2.98 – 2.91 (m, 2H), 2.53 (q, <i>J</i> = 7.4 Hz, 2H), 1.11 (t, <i>J</i> = 7.5 Hz, 3H)
B8	25IP-NBOMe HCl	98.59	500 MHz DMSO- <i>d</i> ₆ δ 9.22 (br s, 2H), 7.50 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.3, 7.4, 1.7 Hz, 1H), 7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H), 6.99 (td, <i>J</i> = 7.4, 1.0 Hz, 1H), 6.80 (s, 1H), 6.79 (s, 1H), 4.12 (s, 2H), 3.83 (s, 3H), 3.73 (s, 3H), 3.73 (s, 3H), 3.21 (hept, <i>J</i> = 6.9 Hz, 1H), 3.07 – 2.99 (m, 2H), 2.98 – 2.91 (m, 2H), 1.15 (d, <i>J</i> = 6.9 Hz, 6H)
B9	25G-NBOMe HCl	96.11	500 MHz DMSO- <i>d</i> ₆ δ 9.29 (br s, 2H), 7.52 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H), 7.09 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H), 7.00 (td, <i>J</i> = 7.5, 1.0 Hz, 1H), 6.66 (s, 1H), 4.14 (s, 2H), 3.84 (s, 3H), 3.73 (s, 3H), 3.58 (s, 3H), 3.11 – 3.02 (m, 2H), 3.05 – 2.95 (m, 2H), 2.12 (s, 3H), 2.04 (s, 3H)
B10	M-NBOMe HCl	96.07	500 MHz DMSO- <i>d</i> ₆ 9.33 (br s, 2H), 7.53 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H), 7.41 (ddd, <i>J</i> = 8.2, 7.4, 1.7 Hz, 1H), 7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H), 6.99 (td, <i>J</i> = 7.5, 1.0 Hz, 1H), 6.55 (s, 2H), 4.11 (s, 2H), 3.83 (s, 3H), 3.76 (s, 6H), 3.62 (s, 3H), 3.16 – 3.08 (m, 2H), 3.00 – 2.93 (m, 2H)
IM1	25H-NBOMe imine soln	> 95 ^e	500 MHz DMSO- <i>d</i> ₆ δ 8.56 (s, 1H), 7.82 (dd, <i>J</i> = 7.7, 1.8 Hz, 1H), 7.42 (ddd, <i>J</i> = 8.3, 7.3, 1.8 Hz, 1H), 7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H), 6.98 (tt, <i>J</i> = 7.5, 0.8 Hz, 1H), 6.87 (d, <i>J</i> = 8.8 Hz, 1H), 6.78 (d, <i>J</i> = 3.1 Hz, 1H), 6.72 (dd, <i>J</i> = 8.8, 3.1 Hz, 1H), 3.82 (s, 3H), 3.74 (s, 3H), 3.75 – 3.68 (m, 2H), 3.65 (s, 3H), 2.84 (t, <i>J</i> = 7.5 Hz, 2H)
IM2	25I-NBOMe imine soln	> 95 ^e	500 MHz DMSO- <i>d</i> ₆ δ 8.54 (s, 1H), 7.82 (dd, <i>J</i> = 7.7, 1.8 Hz, 1H), 7.42 (ddd, <i>J</i> = 8.3, 7.3, 1.8 Hz, 1H), 7.28 (s, 1H), 7.08 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H), 6.97 (tt, <i>J</i> = 7.5, 0.8 Hz, 1H), 6.87 (s, 1H), 3.82 (s, 3H), 3.75 (s, 3H), 3.76 – 3.69 (m, 2H), 3.69 (s, 3H), 2.85 (t, <i>J</i> = 7.3 Hz, 2H)

^a Formulated as the aminium salt except where noted.

^b The stated formulation of the analyte, where provided: C, conflicting claims made on analyte packaging and supporting documentation; HCl, hydrochloride; FB, free base; soln: in methyl acetate solution; N/A, not available; U, unspecified.

^c The per cent ratio of the total analyte integral over the total spectrum integral as reported by Mestrelab's MestReNova software except where noted.

^d br, broad; d, doublet; h, hextet; hept, heptet; obs, partially obscured; m, multiplet; q, quartet; s, singlet; t, triplet; vb, very broad; *J*, coupling constant; Ph, phenyl.

^e Purity as labelled by vendor.

^f Converted to the hydrochloride salt from the supplied free base oil following Shulgin and Shulgin,^[3] recrystallized in oxolane (tetrahydrofuran).

^g A second spectrum acquired in DMF-*d*₇ resolved a multiplet partially obscured in the original spectrum acquired in DMSO-*d*₆.

^h Alleged and sold as 2C-C, a claim unsupported by the spectrum which strongly suggests 2C-E.

ⁱ Formulated as the free base.




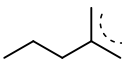

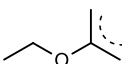
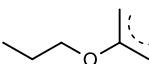
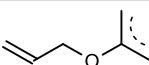
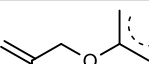

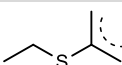
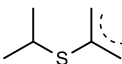
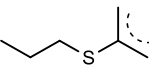
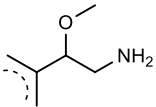
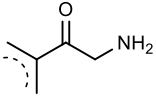
Table 3: Multiplets for phenylethylamine, *N*-benzyl phenylethylamine, and *N*-benzylidene phenylethylamine scaffold moieties

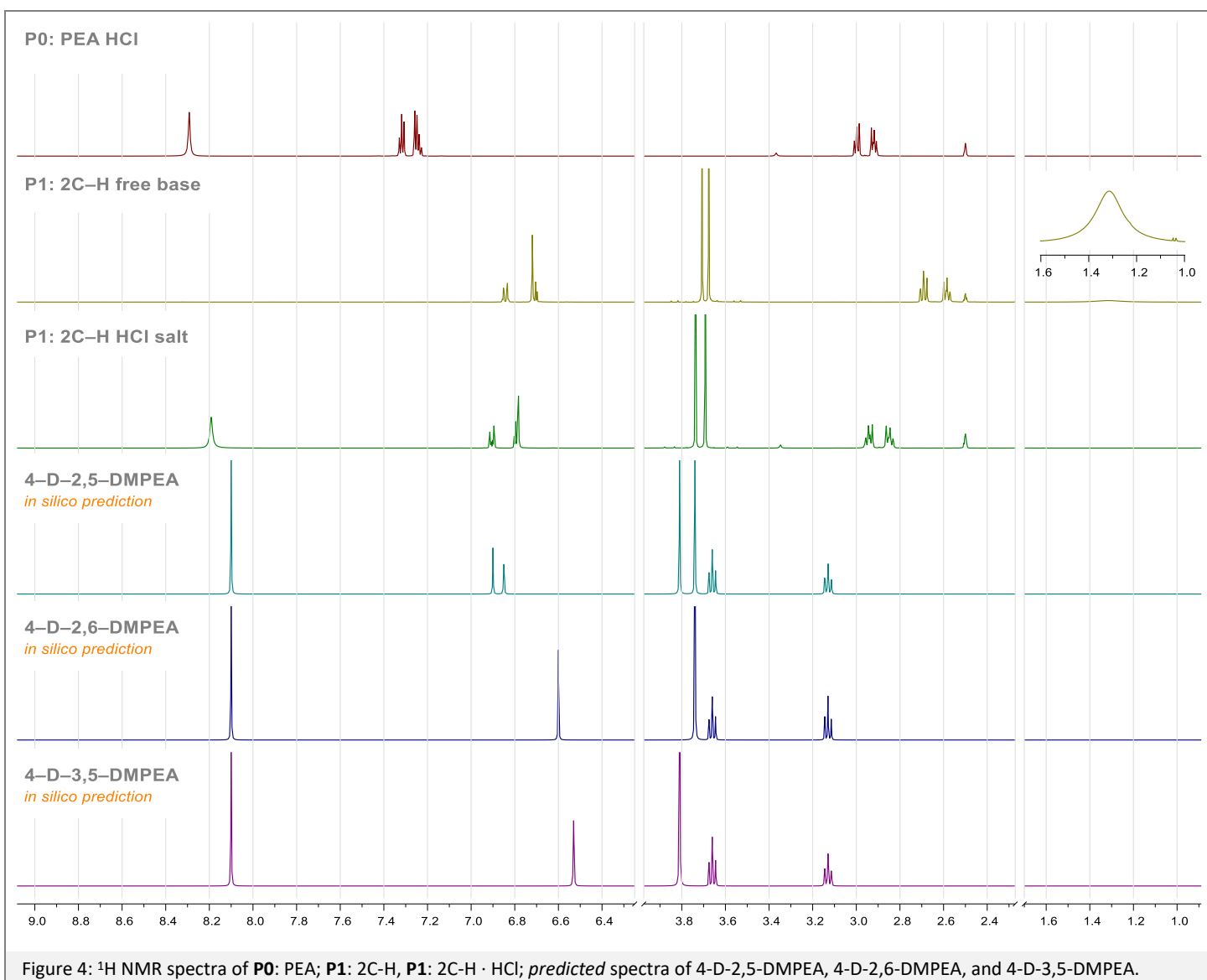
Analyte	$NH_2^+ NH_3^+ 2'-OH$	2-H 3-H 6-H	2-OCH ₃ 3-OCH ₃ 5-OCH ₃ 2'-OCH ₃	$\alpha-H_2 \beta-H_2$	NH NH ₂	
P0 PEA · HCl	8.29 (br s, 3H)	7.34 – 7.21 (m, 5H)		3.02 – 2.97 (m, 2H), 2.95 – 2.89 (m, 2H)		
P1 2C-H base		6.88 – 6.81 (m, 1H),	6.76 – 6.68 (m, 2H)	3.71 (s, 3H), 3.68 (s, 3H)	2.73 – 2.66 (m, 2H), 2.63 – 2.55 (m, 2H)	1.32 (vb s, 2H)
P1 2C-H · HCl	8.19 (br s, 3H)	6.94 – 6.87 (m, 1H),	6.82 – 6.76 (m, 2H)	3.74 (s, 3H), 3.69 (s, 3H)	2.98 – 2.91 (m, 2H), 2.88 – 2.81 (m, 2H)	
P2 2C-D	8.15 (br s, 3H)	6.82 (d, $J = 0.8$ Hz, 1H),	6.78 (s, 1H)	3.73 (s, 6H)	2.97 – 2.89 (m, 2H), 2.87 – 2.80 (m, 2H)	
P3 2C-E	8.02 (br s, 3H)	6.81 (s, 1H),	6.79 (s, 1H)	3.74 (s, 3H), 3.74 (s, 3H)	3.00 – 2.91 (m, 2H), 2.86 – 2.79 (m, 2H)	
P4 2C-P	8.04 (br s, 3H)	6.79 (s, 2H)		3.74 (s, 3H), 3.73 (s, 3H)	2.99 – 2.90 (m, 2H), 2.86 – 2.79 (m, 2H)	
P4 2C-P in DMF- <i>d</i> ₇	8.64 (br s, 3H)	6.99 (s, 1H),	6.87 (s, 1H)	3.82 (s, 3H), 3.80 (s, 3H)	3.26 – 3.19 (m, 2H), 3.10 – 3.03 (m, 2H)	
P5 2C-IP	8.00 (br s, 3H)	6.81 (s, 1H),	6.79 (s, 1H)	3.76 (s, 3H), 3.74 (s, 3H)	2.99 – 2.92 (m, 2H), 2.85 – 2.78 (m, 2H)	
P6a 2C-E	8.03 (br s, 3H)	6.81 (s, 1H),	6.79 (s, 1H)	3.74 (s, 3H), 3.74 (s, 3H)	2.98 – 2.91 (m, 2H), 2.86 – 2.79 (m, 2H)	
P6b 2C-C	8.18 (br s, 3H)	7.07 (s, 1H),	7.05 (s, 1H)	3.80 (s, 3H), 3.76 (s, 3H)	3.00 – 2.93 (m, 2H), 2.90 – 2.83 (m, 2H)	
P7 2C-I	7.99 (br s, 3H)	7.33 (s, 1H),	6.90 (s, 1H)	3.77 (s, 3H), 3.76 (s, 3H)	3.00 – 2.93 (m, 2H), 2.87 – 2.80 (m, 2H)	
P8 2C-T	7.97 (br s, 3H)	6.83 (s, 1H),	6.76 (s, 1H)	3.80 (s, 3H), 3.76 (s, 3H)	2.98 – 2.91 (m, 2H), 2.86 – 2.79 (m, 2H)	
P9 2C-T-2	7.98 (br s, 3H)	6.85 (s, 1H),	6.83 (s, 1H)	3.78 (s, 3H), 3.76 (s, 3H)	3.01 – 2.92 (m, 2H), 2.86 – 2.79 (m, 2H)	
P10 2C-T-4	7.98 (br s, 3H)	6.91 (s, 1H),	6.87 (s, 1H)	3.77 (s, 3H), 3.76 (s, 3H)	3.01 – 2.93 (m, 2H), 2.87 – 2.80 (m, 2H)	
P11 2C-T-7	8.02 (br s, 3H)	6.85 (s, 1H),	6.82 (s, 1H)	3.77 (s, 3H), 3.76 (s, 3H)	2.99 – 2.92 (m, 2H), 2.86 – 2.79 (m, 2H)	
P12 Escaline	8.22 (br s, 3H)	6.56 (s, 2H)		3.75 (s, 6H)	3.06 – 2.99 (m, 2H), 2.89 – 2.81 (m, 2H)	
P13a Proscaline	8.05 (br s, 3H)	6.56 (s, 2H)		3.76 (s, 6H)	3.07 – 3.00 (m, 2H), 2.85 – 2.78 (m, 2H)	
P13b Proscaline	8.20 (br s, 3H)	6.56 (s, 2H)		3.75 (s, 6H)	3.06 – 2.98 (m, 2H), 2.89 – 2.80 (m, 2H)	
P14 AL	8.13 (br s, 3H)	6.57 (s, 2H)		3.76 (s, 6H)	3.06 – 2.99 (m, 2H), 2.87 – 2.80 (m, 2H)	
P15 MAL	8.03 (br s, 3H)	6.56 (s, 2H)		3.76 (s, 6H)	3.06 – 3.00 (m, 2H), 2.86 – 2.78 (m, 2H)	
P16 BOD	8.14 (br s, 3H)	6.91 (d, $J = 0.9$ Hz, 1H),	6.82 (s, 1H)	3.75 (s, 3H), 3.75 (s, 3H)	2.96 – 2.82 (m, 2H)	
P17 bk-2C-B	8.28 (br s, 3H)	7.59 (s, 1H),	7.43 (s, 1H)	3.94 (s, 3H), 3.86 (s, 3H)		
B1 25B-NBOMe	9.18 (br s, 2H)	7.19 (s, 1H),	7.02 (s, 1H)	3.83 (s, 3H), 3.79 (s, 3H), 3.75 (s, 3H)	3.10 – 3.03 (m, 2H), 3.00 – 2.93 (m, 2H)	
B2 25C-NBOMe	9.23 (br s, 2H)	7.08 (s, 1H),	7.04 (s, 1H)	3.83 (s, 3H), 3.80 (s, 3H), 3.74 (s, 3H)	3.10 – 3.01 (m, 2H), 3.03 – 2.94 (m, 2H)	
B3 25I-NBOMe	9.15 (br s, 2H)	7.32 (s, 1H),	6.90 (s, 1H)	3.83 (s, 3H), 3.77 (s, 3H), 3.73 (s, 2H)	3.09 – 3.00 (m, 2H), 2.99 – 2.92 (m, 2H)	
B4 25I-NBOH	9.22 (vb s, 3H)	7.32 (s, 1H),	6.89 (s, 1H)	3.77 (s, 3H), 3.74 (s, 3H)	3.09 – 3.02 (m, 2H), 3.00 – 2.91 (m, 2H)	
B5 25I-NBMD		7.25 (s, 1H),	6.84 (s, 1H)	3.73 (s, 3H), 3.71 (d, $J = 0.3$ Hz, 3H)	2.69 – 2.65 (m, 4H)	1.98 (br s, 1H)
B6 25D-NBOMe	9.18 (br s, 2H)	6.88 – 6.79 (m, 1H),	6.77 (s, 1H)	3.83 (s, 3H), 3.73 (s, 3H), 3.71 (s, 3H)	3.10 – 2.96 (m, 2H), 3.00 – 2.88 (m, 2H)	
B7 25E-NBOMe	9.20 (br s, 2H)	6.80 (s, 1H),	6.78 (s, 1H)	3.83 (s, 3H), 3.73 (s, 3H), 3.72 (s, 3H)	3.07 – 2.99 (m, 2H), 2.98 – 2.91 (m, 2H)	
B8 25IP-NBOMe	9.22 (br s, 2H)	6.80 (s, 1H),	6.79 (s, 1H)	3.83 (s, 3H), 3.73 (s, 3H), 3.73 (s, 3H)	3.07 – 2.99 (m, 2H), 2.98 – 2.91 (m, 2H)	
B9 25G-NBOMe	9.29 (br s, 2H)	6.66 (s, 1H)		3.84 (s, 3H), 3.73 (s, 3H), 3.58 (s, 3H)	3.11 – 3.02 (m, 2H), 3.05 – 2.95 (m, 2H)	
B10 M-NBOMe	9.33 (br s, 2H)	6.55 (s, 2H)		3.83 (s, 3H), 3.76 (s, 6H)	3.16 – 3.08 (m, 2H), 3.00 – 2.93 (m, 2H)	
IM1 25H-NBOMe imine		6.87 (d, $J = 8.8$ Hz, 1H), 6.72 (dd, $J = 8.8, 3.1$ Hz, 1H)	6.78 (d, $J = 3.1$ Hz, 1H)	3.82 (s, 3H), 3.74 (s, 3H), 3.65 (s, 3H)	3.75 – 3.68 (m, 2H), 2.84 (t, $J = 7.5$ Hz, 2H)	
IM2 25I-NBOMe imine		7.28 (s, 1H),	6.87 (s, 1H)	3.82 (s, 3H), 3.75 (s, 3H), 3.69 (s, 3H)	3.76 – 3.69 (m, 2H), 2.85 (t, $J = 7.3$ Hz, 2H)	

Table 4: Multiplets for *N*-benzyl phenylethanimine and *N*-benzylidene phenylethanimine scaffold moieties

Analyte		3'-H 4'-H 5'-H 6'-H				2'-O-CH ₂ -O-3'	α'-H ₂
B1	25B-NBOMe	7.49 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H),	7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H),	7.00 (td, <i>J</i> = 7.4, 1.1 Hz, 1H)		4.11 (s, 2H)
B2	25C-NBOMe	7.50 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.2, 7.4, 1.7 Hz, 1H),	7.14 – 7.05 (m, 1H),	7.00 (td, <i>J</i> = 7.5, 1.1 Hz, 1H)		4.11 (s, 2H)
B3	25I-NBOMe	7.48 (dd, <i>J</i> = 7.5, 1.8 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H),	7.08 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H),	6.99 (td, <i>J</i> = 7.4, 1.1 Hz, 1H)		4.11 (s, 2H)
B4	25I-NBOH	7.39 (dd, <i>J</i> = 7.6, 1.7 Hz, 1H),	7.23 (ddd, <i>J</i> = 8.1, 7.3, 1.7 Hz, 1H),	6.97 (dd, <i>J</i> = 8.2, 1.1 Hz, 1H),	6.84 (td, <i>J</i> = 7.4, 1.1 Hz, 1H)		4.09 (s, 2H)
B5	25I-NBMD			6.86 – 6.78 (m, 1H),	6.81 – 6.72 (m, 2H)	5.96 (s, 2H)	3.66 (d, <i>J</i> = 0.6 Hz, 2H)
B6	25D-NBOMe	7.49 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H),	7.08 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H),	7.00 (td, <i>J</i> = 7.5, 1.1 Hz, 1H)		4.11 (s, 2H)
B7	25E-NBOMe	7.49 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H),	7.09 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H),	7.00 (td, <i>J</i> = 7.5, 1.0 Hz, 1H)		4.12 (s, 2H)
B8	25IP-NBOMe	7.50 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.3, 7.4, 1.7 Hz, 1H),	7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H),	6.99 (td, <i>J</i> = 7.4, 1.0 Hz, 1H)		4.12 (s, 2H)
B9	25G-NBOMe	7.52 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.3, 7.5, 1.7 Hz, 1H),	7.09 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H),	7.00 (td, <i>J</i> = 7.5, 1.0 Hz, 1H)		4.14 (s, 2H)
B10	M-NBOMe	7.53 (dd, <i>J</i> = 7.5, 1.7 Hz, 1H),	7.41 (ddd, <i>J</i> = 8.2, 7.4, 1.7 Hz, 1H),	7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H),	6.99 (td, <i>J</i> = 7.5, 1.0 Hz, 1H)		4.11 (s, 2H)
IM1	25H-NBOMe imine	7.82 (dd, <i>J</i> = 7.7, 1.8 Hz, 1H),	7.42 (ddd, <i>J</i> = 8.3, 7.3, 1.8 Hz, 1H),	7.08 (dd, <i>J</i> = 8.4, 1.0 Hz, 1H),	6.98 (tt, <i>J</i> = 7.5, 0.8 Hz, 1H)		
IM2	25I-NBOMe imine	7.82 (dd, <i>J</i> = 7.7, 1.8 Hz, 1H),	7.42 (ddd, <i>J</i> = 8.3, 7.3, 1.8 Hz, 1H),	7.08 (dd, <i>J</i> = 8.3, 1.0 Hz, 1H),	6.97 (tt, <i>J</i> = 7.5, 0.8 Hz, 1H)		

Table 5: Multiplets for phenyl 4-substituents, β -substituents, and other atypical moieties. Preferred IUPAC names precede · deprecated names.

4-methyl 	P2: 2C-D P16: BOD	CH_3 - Ph 2.13 (d, $J = 0.7$ Hz, 3H) 2.16 (d, $J = 0.9$ Hz, 3H)	CH_3 - Ph B6: 25D-NBOMe B9: 25G-NBOMe	CH_3 - Ph 2.13 (d, $J = 0.7$ Hz, 3H) 2.04 (s, 3H), 2.12 (s, 3H)
4-ethyl 	P3: 2C-E B7: 25E-NBOMe	CH_3 - CH_2 - Ph 1.11 (t, $J = 7.5$ Hz, 3H) 1.11 (t, $J = 7.5$ Hz, 3H)	CH_3 - CH_2 - Ph 2.54 (q, $J = 7.5$ Hz, 2H) 2.53 (q, $J = 7.4$ Hz, 2H)	
4-propan-2-yl · isopropyl 	P5: 2C-IP B8: 25IP-NBOMe	$(CH_3)_2$ - CH - Ph 1.15 (d, $J = 6.9$ Hz, 6H) 1.15 (d, $J = 6.9$ Hz, 6H)	$(CH_3)_2$ - CH - Ph 3.21 (hept, $J = 6.9$ Hz, 1H) 3.21 (hept, $J = 6.9$ Hz, 1H)	
4-propyl 	P4: 2C-P P4: 2C-P in DMF- d_7	CH_3 - CH_2 - CH_2 - Ph 0.89 (t, $J = 7.4$ Hz, 3H) 0.92 (t, $J = 7.4$ Hz, 3H)	CH_3 - CH_2 - CH_2 - Ph 1.60 – 1.45 (m, 2H) 1.63 – 1.52 (m, 2H)	CH_3 - CH_2 - CH_2 - Ph 2.51 – 2.45 (m, 2H) 2.59 – 2.52 (m, 2H)
4-methoxy 	B10: M-NBOMe	CH_3 - O - Ph 3.62 (s, 3H)		
4-ethoxy 	P12: Escaline	CH_3 - CH_2 - O - Ph 1.20 (t, $J = 7.0$ Hz, 3H)	CH_3 - CH_2 - O - Ph 3.85 (q, $J = 7.0$ Hz, 2H)	
4-propoxy 	P13a: Proscaline P13b: Proscaline	CH_3 - CH_2 - CH_2 - O - Ph 0.94 (t, $J = 7.4$ Hz, 3H) 0.93 (t, $J = 7.4$ Hz, 3H)	CH_3 - CH_2 - CH_2 - O - Ph 1.61 (h, $J = 7.1$ Hz, 2H) 1.66 – 1.55 (m, 2H)	CH_3 - CH_2 - CH_2 - O - Ph 3.79 – 3.72 (m, 2H) obs. 3.80 – 3.71 (m, 2H) obs.
4-(prop-2-en-1-yl)oxy · allyloxy 	P14: AL	$CH_2 = CH$ - CH_2 - O - Ph 5.28 (dq, $J = 17.3, 1.8$ Hz, 1H) 5.17 – 5.10 (m, 1H)	$CH_2 = CH$ - CH_2 - O - Ph 5.98 (ddt, $J = 17.3, 10.4, 5.6$ Hz, 1H)	$CH_2 = CH$ - CH_2 - O - Ph 4.35 (dt, $J = 5.6, 1.6$ Hz, 2H)
4-(2-methylprop-2-en-1-yl)oxy · methallyloxy 	P15: MAL	$CH_2 = C(CH_3)$ - CH_2 - O - Ph 5.02 – 4.97 (m, 1H) 4.89 – 4.83 (m, 1H)	$CH_2 = C(CH_3)$ - CH_2 - O - Ph 1.81 – 1.76 (m, 3H)	$CH_2 = C(CH_3)$ - CH_2 - O - Ph 4.27 – 4.22 (m, 2H)
4-methylsulfanyl · methylthio 	P8: 2C-T	CH_3 - S - Ph 2.41 (s, 3H)		
4-ethylsulfanyl · ethylthio 	P9: 2C-T-2	CH_3 - CH_2 - S - Ph 1.23 (t, $J = 7.3$ Hz, 3H)	CH_3 - CH_2 - S - Ph 2.92 (q, $J = 7.3$ Hz, 2H)	
4-(propan-2-yl)sulfanyl · isopropylthio 	P10: 2C-T-4	$(CH_3)_2$ - CH - S - Ph 1.21 (d, $J = 6.6$ Hz, 6H)	$(CH_3)_2$ - CH - S - Ph 3.54 (hept, $J = 6.6$ Hz, 1H)	
4-propylsulfanyl · propylthio 	P11: 2C-T-7	CH_3 - CH_2 - CH_2 - S - Ph 0.98 (t, $J = 7.3$ Hz, 3H)	CH_3 - CH_2 - CH_2 - S - Ph 1.58 (h, $J = 7.3$ Hz, 2H)	CH_3 - CH_2 - CH_2 - S - Ph 2.88 (t, $J = 7.1$ Hz, 2H)
β -methoxy 	P16: BOD	β - H 4.78 (dd, $J = 8.9, 3.9$ Hz, 1H)	β - O - CH_3 3.21 (s, 3H)	α - H_2 2.96 – 2.82 (m, 2H)
β -oxo · keto 	P17: bk-2C-B	α - H_2 4.31 (q, $J = 5.2$ Hz, 2H)		



Analysis

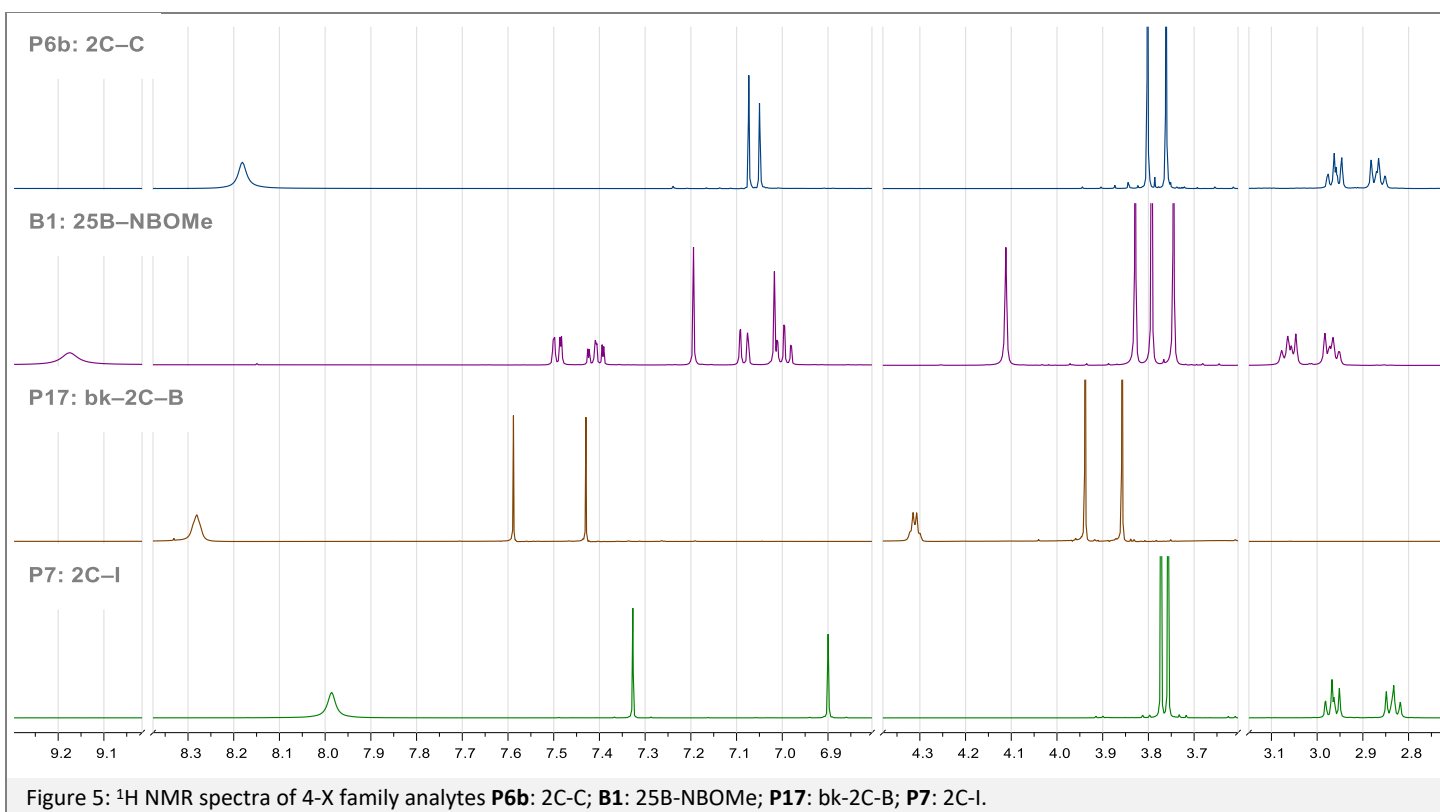
Almost all of the analytes appear to be formulated as simple aminium salts e.g. a hydrochloride, facilitating comparison. The protonated aminium resonance typically appears as a broad singlet, near or downfield of 8 ppm in primary amines; 9.15 ppm in secondary amines. The unprotonated amine group found in the two free base analytes resonates as a very broad singlet upfield of 2 ppm (cf. **P1**: 2C-H · HCl vs. 2C-H in Figure 4). The quintet at 2.50 ppm is characteristic of trace amounts of the partially deuterated solvent DMSO- d_5 .

The phenyl ring substitution patterns most often encountered are the 2,5- and 3,5-dimethoxy with *para*-substitution. Though purported TMA-6¹ has occasionally been seen, the 2,6-dimethoxy pattern appears largely absent from the grey market.

A 4-substituted 2,5-dimethoxy spectrum typically exhibits a discrete 3H singlet for each methoxy substituent and a discrete 1H singlet for each unsubstituted aryl hydrogen. The methoxy groups are not magnetically equivalent, nor are the aryl hydrogens, and each may resonate at a discrete shift. In some cases, these singlet resonances may overlap or appear isochronous, which has particularly been reported in the earlier literature.

In a 4-substituted 3,5- or 2,6-dimethoxy spectrum, both methoxy substituents are magnetically equivalent by symmetry, as are both aryl hydrogens. The methoxy groups appear as a single 6H singlet, the aryl hydrogen as a 2H singlet. Figure 4 shows *predicted* spectra for 2-(4-deuterophenyl)ethan-1-amine with 2,5-, 3,5-, and 2,6-dimethoxy substitution. The 4-deutero substituent functions as a placeholder keeping the simulation free of spurious peaks arising from *ortho*- and *meta*-coupling of the aryl hydrogen.

¹ TMA-6: 1-(2,4,6-trimethoxyphenyl)propan-2-amine.



¹H NMR spectra have been reported for 2C-H in DMSO-*d*₆,^[29] in CDCl₃,^[15,30] and in D₂O,^[31–33] and for the NBOMe analogue 25H-NBOMe¹ in DMSO-*d*₆.^[29] Shaler and Padden have compared ¹H NMR spectra of the base and the hydrobromide salt of the α-methylated homologue² in CDCl₃.^[34] ¹H NMR spectra have been reported for the *N*,α-dimethyl homologues (i.e. the methamphetamine analogues) of 13 mono-, di-, and trimethoxyphenylethanamines in CDCl₃.^[35]

The 4-X family: 4-halo-2,5-dimethoxyphenyl pattern

Members of the 4-X family share a 4-halo-2,5-dimethoxyphenyl substitution pattern, with a bromo, chloro or iodo *para*-substituent. Shulgin reported the 4-fluoro analogue³ largely inactive even at 250 mg, an order of magnitude less potent than of the other three.^[3] Alleged 2C-F has been advertised on the grey market lately. None has been obtained and its actual composition is unknown.

Regulatory hurdles precluded obtaining the controlled 4-bromo analogue, 2C-B.^[36] However, the 2C-B ring substitution pattern is present in other, more recent arrivals to the grey market. Samples alleged to be 25B-NBOMe and bk-2C-B, uncontrolled analogues of 2C-B, were purchased instead.

¹ 25H-NBOMe: 2-(2,5-dimethoxyphenyl)-*N*-[(2-methoxyphenyl)methyl]ethan-1-amine.

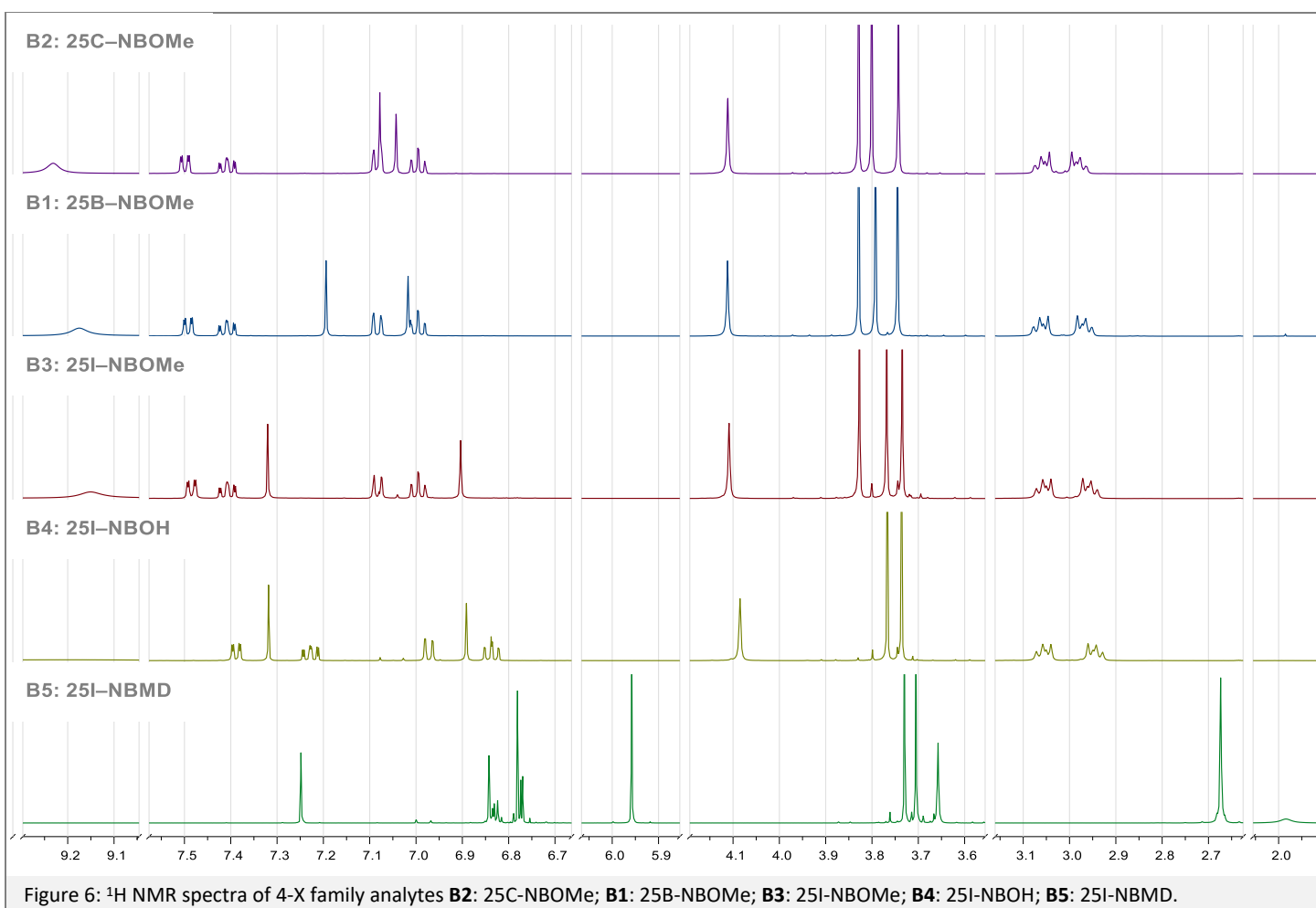
² 2,5-DMA: 1-(2,5-dimethoxyphenyl)propan-2-amine. The ambiguous code DMA may refer to 2,5-DMA or 3,4-DMA.

³ 2C-F: 2-(4-fluoro-2,5-dimethoxyphenyl)ethan-1-amine.

The 4-X spectra in Figures 5 and 6 show a clear correlation between the 4-halo substituent and the shift of the two phenyl hydrogen. Table 6 compares our findings (shaded rows) with reported data acquired in DMSO-*d*₆. The consistency of the shift separation ($\Delta\delta$) suggests a facile method for differentiating the 4-halo substituent of 4-X family compounds.

Table 6: Phenyl peak shift separation in ppm for 4-X family compounds

	Compound	Source	← δ	δ →	Δ δ
Chloro	2C-C	P6b	7.07	7.05	0.02
	25C-NBOMe	B2	7.08	7.04	0.04
	25C-NBOMe	Hansen et al. ^[37: 3a]	7.08	7.04	0.04
Bromo	2C-B	Martins ^[29: 8]	7.21	7.00	0.21
	bk-2C-B	P17	7.59	7.43	0.16
	bk-2C-B	Power et al. ^[38: 3]	7.60	7.44	0.16
	DOB	Heim ^[15: 35]	7.21	7.04	0.17
	DOB	Martins ^[29: 9]	7.20	7.01	0.19
	25B-NBOMe	B1	7.19	7.02	0.17
	25B-NBOMe	Hansen et al. ^[37: 2a]	7.17	7.01	0.16
Iodo	2C-I	P7	7.33	6.90	0.43
	25I-NBOMe	B3	7.32	6.90	0.42
	25I-NBOMe	Hansen et al. ^[37: 1a]	7.32	6.91	0.41
	25I-NBOH	B4	7.32	6.89	0.43
	25I-NBOH	Hansen et al. ^[37: 1b]	7.30	6.88	0.42
	25I-NBOH	Heim ^[15: 235]	7.32	6.90	0.42
	25I-NBMD	B5	7.25	6.84	0.41
	25I-NBMD	Hansen et al. ^[37: 1d]	7.30	6.88	0.42
	DOI	Heim ^[15: 36]	7.34	6.91	0.43
	25I-NBOMe imine	IM2	7.28	6.87	0.41



P17: bk-2C-B is the only β -oxo¹ phenylethanamine we have seen and our only analyte of this kind. In contrast, a plethora of grey-market β -oxo-phenylpropamines and higher homologues (“cathinones”) are on offer, typically as secondary or cyclic amines.

The effects of the β -oxo group is readily seen in Figure 5. The highly conserved pair of 2H α - and β -multiplets seen in the other three spectra are replaced by a 2H pseudo-quartet from the α methylene protons, while the 3H aminium peak shows a trace of triplet symmetry. This suggests possible coupling, as reported in an unattributed analysis of ^1H , ^{13}C , and 2-D spectra.^[39] A similar pseudo-quartet appears in the supplementary data of the analysis by Power et al.^[38] The aryl methoxy peaks are shifted noticeably downfield and the aryl hydrogen peaks even more so though their separation is not appreciably effected (see Table 6).

^1H NMR spectra of two additional analogues of 25I-NBOMe are shown in Figure 6. The NBOMe moiety is replaced by a 2-hydroxybenzyl group (NBOH) in analyte **B4**: 25I-NBOH, and by

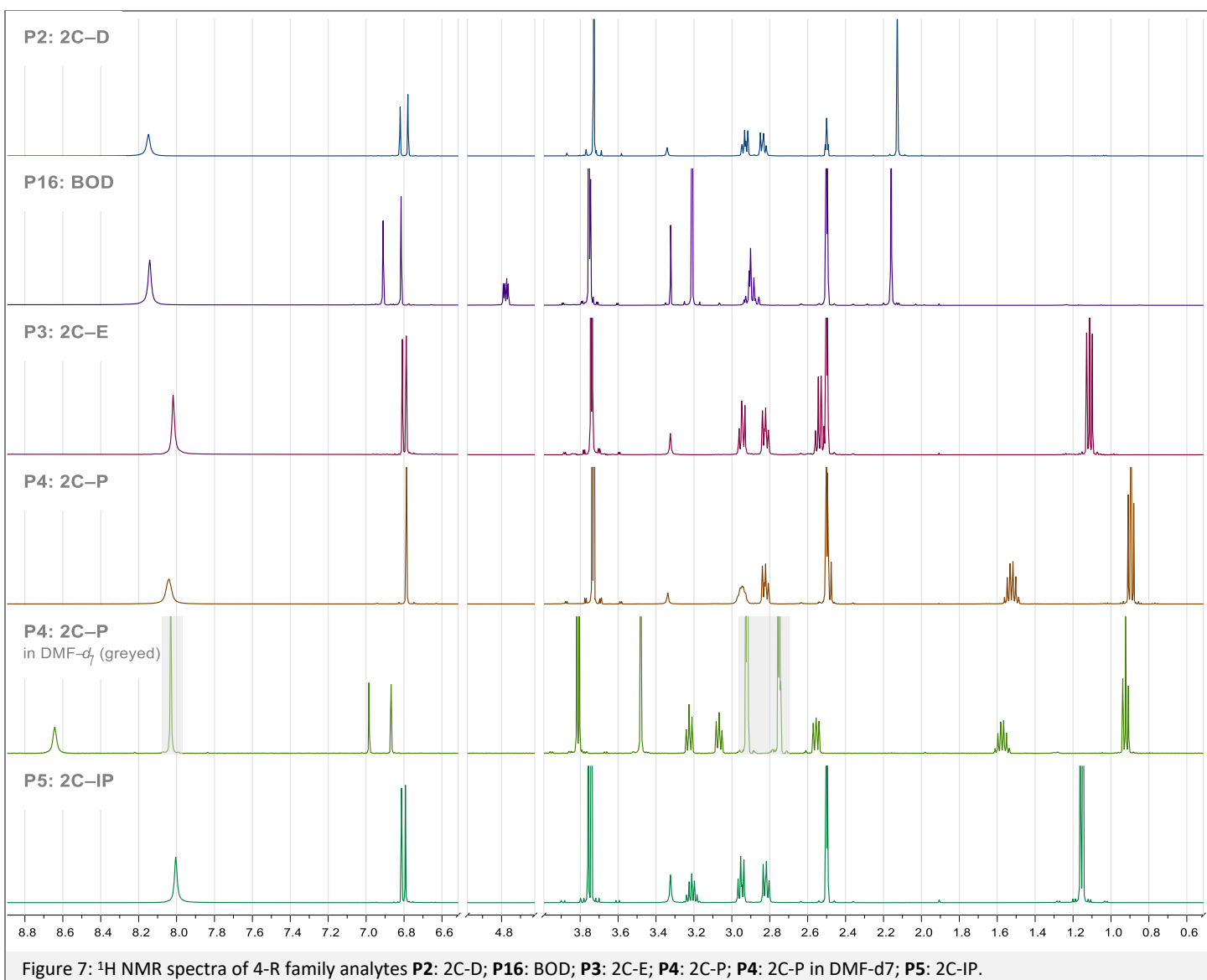
2H-(1,3-benzodioxol-4-yl)methyl group (NBMD) in analyte **B5**: 25I-NBMD.²

Comparing the **B3**: 25I-NBOMe and **B4**: 25I-NBOH spectra, in the latter only two 3H singlet methoxy peaks are present, reflecting the loss of the benzyl methoxy group; the characteristic multiplet pattern of the four benzyl protons is shifted upfield about 0.13 ppm on average; and in place of a broad 2H aminium peak at 9.15 ppm we found a very broad 3H peak spanning ca. 2 ppm at 9.22 ppm (not discernable in Figure 6; see Table 3). This signal may result from intramolecular hydrogen bonding and proton exchange between the phenolic *ortho*-hydroxyl group of the NBOH moiety and the protonated amine, perhaps facilitated by the hydrogen-bond accepting solvent. However, both Heim^[15] and Hansen^[16,37] report a discrete and sharp phenolic peak in DMSO-*d*₆.

The spectrum of analyte **B5**: 25I-NBMD is markedly different from its NBOMe and NBOH analogues. It appears to be the only *N*-benzyl analyte formulated as the free amine rather than an aminium salt, consistent with the presence of a 1H upfield peak

¹ Though the prefix *bk*- for *beta*- or β -*keto* is widely used, *oxo* is now the preferred IUPAC prefix and *keto* is deprecated.

² NBMD stems from *N*-BenzylMethyleneDioxy. IUPAC has ruled *methylenedioxy* unacceptably ambiguous (is it $-\text{OCH}_2\text{O}-$ or $-\text{CH}_2\text{OO}-$?) and deprecated.



at 1.98 ppm, the absence of any 2H downfield peak, and the substantial upfield shift of the entire spectrum (cf. **P1** as the free base vs. the hydrochloride salt in Figure 4). Also, the α - and β -methylene groups have coalesced into a 4H multiplet, the familiar splitting pattern of the NBOMe/NBOH benzyl protons is absent, and a 2H singlet from the benzodioxole methylene group is seen at 5.96 ppm.

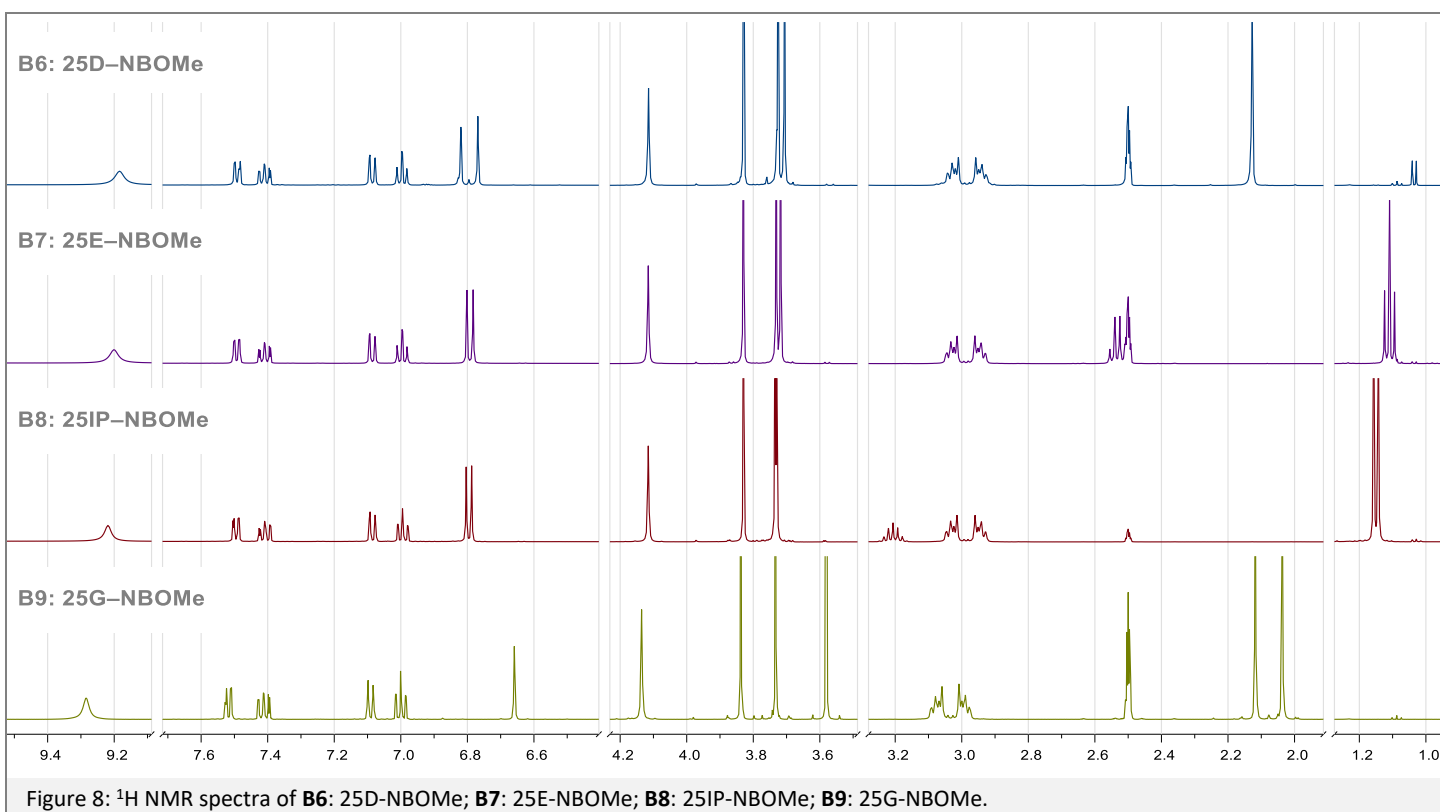
^1H NMR spectra have been reported for 2C-B in $\text{DMSO-}d_6$,^[29] in CDCl_3 ,^[15,40] in CD_3OD ,^[41–43] and in D_2O ,^[31,32,42,44]; for 2C-C in CDCl_3 ^[45,46] and in D_2O ,^[31,46]; for 2C-I in CDCl_3 ^[43] and in D_2O ,^[31,32,47,48]; for bk-2C-B in $\text{DMSO-}d_6$,^[38,39] and in CDCl_3 .^[49]

^1H NMR spectra have been reported for 25B-NBOMe in $\text{DMSO-}d_6$,^[16,37] and in CDCl_3 ,^[15,50]; for 25C-NBOMe in $\text{DMSO-}d_6$,^[37] in CDCl_3 ,^[51] and in CD_3OD ,^[16,52]; for 25I-NBOMe in $\text{DMSO-}d_6$,^[16,37,53] in CDCl_3 ,^[15,54–56] and in CD_3OD ,^[57]; for 25I-NBOH in $\text{DMSO-}d_6$,^[15,16,37]; for 25I-NBMD in $\text{DMSO-}d_6$.^[16,37]

The 4-R family: 4-alkyl-2,5-dimethoxyphenyl pattern

Five primary amine analytes (Figure 7) and four secondary amine analytes (Figure 8) share the 4-alkyl-2,5-dimethoxyphenyl pattern where the 4-alkyl substituent is methyl (**P2**: 2C-D, **P16**: BOD, **B6**: 25D-NBOMe), ethyl (**P3**: 2C-E, **B7**: 25E-NBOMe), propyl (**P4**: 2C-P), or isopropyl (**P5**: 2C-IP, **B8**: 25IP-NBOMe). Analyte **B9**: 25G-NBOMe is the 3,4-dimethyl homologue of **B6**. Fortunately, the 4-alkyl substitution introduces additional peaks while leaving the core (**P1**: 2C-H) spectrum essentially undisturbed, simplifying analysis.

In analyte **P2**: 2C-D the line broadening of the 4-methyl peak (apparent 3H singlet at 2.13 ppm arises from long range coupling to the *ortho*-hydrogen at ring position 3, which shows similar line broadening. Analogous coupling is seen in the β -methoxy analogue **P16**: BOD, and the NBOMe analogue **B6**: NBOMe. The coupling constants are consistent with the published value of 0.7 Hz.^[58]



In the **P3**: 2C-E spectrum, the classical ethyl group splitting pattern is seen: a 2H quartet and 3H triplet. Similarly, the isopropyl group of **P5**: 2C-IP is readily identified by the characteristic methine proton 1H heptet signal and the 6H doublet signal from the adjacent, isochronous methyl groups.

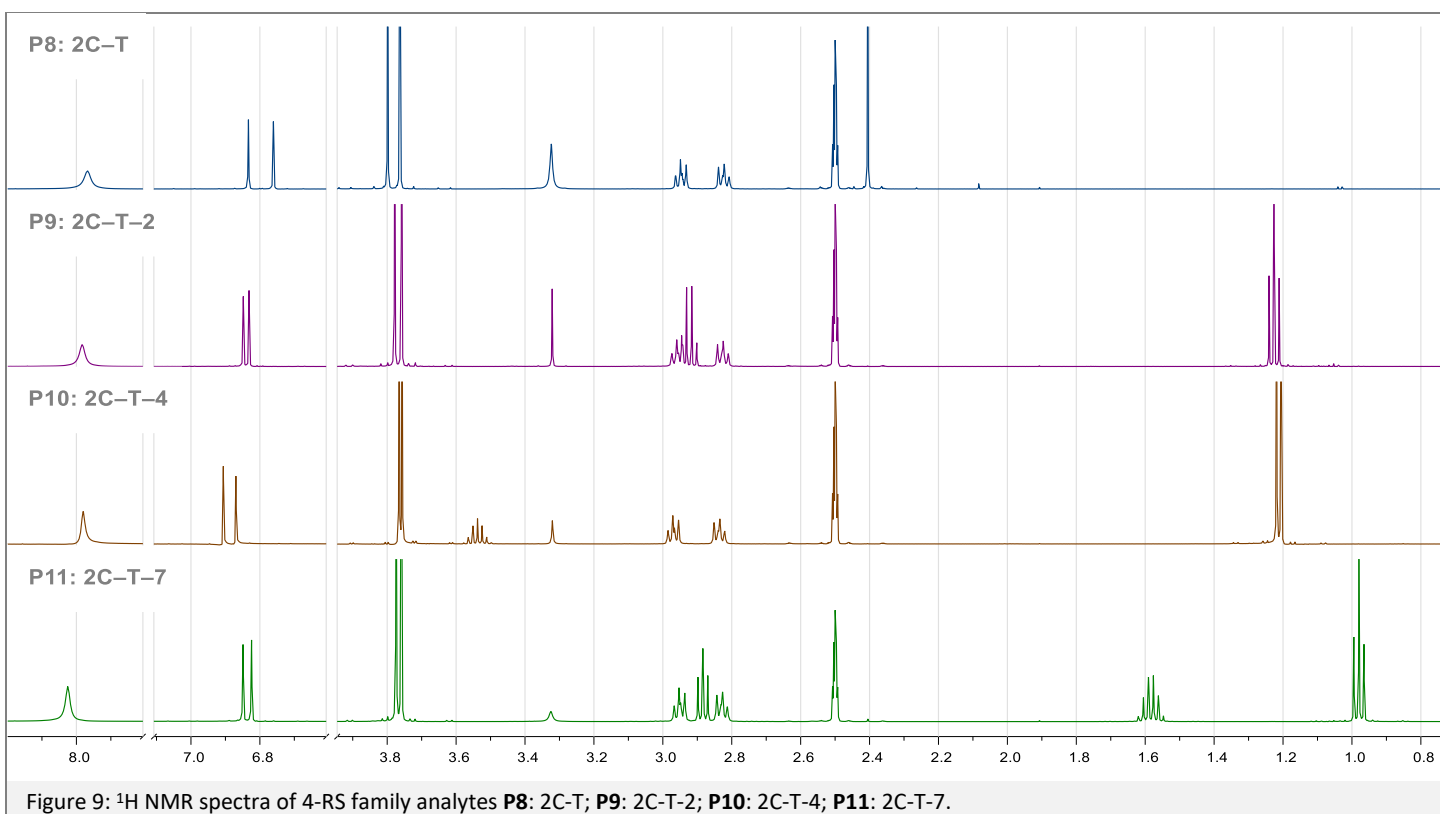
The $\text{DMSO-}d_6$ solvent signal at 2.50 ppm hinders interpretation of the **P4**: 2C-P spectrum by obscuring the expected 2H triplet of the propyl group while the corresponding 2H multiplet and 3H triplet are clearly present. A second spectrum, acquired in deuterated *N,N*-dimethylformamide ($\text{DMF-}d_7$), allows for an unobstructed view of the propyl multiplet.

Interestingly, though the 3H triplet of the terminal methyl moiety appears typical, much like the corresponding peak seen in **P3**: 2C-E, the splitting pattern of the methylene protons is more complex. The methylene group alpha to the ring does not split as a simple first-order triplet, nor does the methylene group beta to the ring split as a first-order hexet. The same divergence from simple first-order splitting is seen in the other propyl-substituted analytes.

^1H NMR spectra have been reported for 2C-D in CDCl_3 ; for the regioisomer iso-2C-D¹ in CDCl_3 and in CD_3OD ^[59]; for 2C-E in $\text{DMSO-}d_6$ ^[60] and in D_2O ^[61]; for 2C-P in D_2O ^[48]; for BOD in CDCl_3 ^[62] and in D_2O ^[63]. We have not found ^1H NMR spectra reported for 2C-IP or 25IP-NBOMe.

^1H NMR spectra have been reported for 25D-NBOMe in $\text{DMSO-}d_6$ ^[16,37,64]; for 25E-NBOMe in $\text{DMSO-}d_6$ ^[16,37,64] and in CDCl_3 ^[65]; for 25G-NBOMe in $\text{DMSO-}d_6$ ^[64]

¹ iso-2C-D: 2-(2,4-dimethoxy-3-methylphenyl)ethan-1-amine.



The 4-RS family: 4-alkylsulfanyl-2,5-dimethoxyphenyl pattern

The 4-RS family analytes (Figure 9) are sulfide¹ analogues of 4-R family compounds, the result of interposing a sulfur atom between the phenyl ring and 4-alkyl substituent. Four of the analytes are members of this family, each a primary amine, where the alkyl substituent is methylsulfanyl (**P8**: 2C-T), ethylsulfanyl (**P9**: 2C-T-2) isopropylsulfanyl (**P10**: 2C-T-4), or propylsulfanyl (**P11**: 2C-T-7).

The splitting pattern arising from a particular alkyl substituent is the same in either family, e.g. 4-ethyl and 4-ethylsulfanyl. However, the alkyl-peak shifts of the 4-RS compounds are further downfield.

We have not seen the corresponding *ether* analogous on the grey market, the 4-alkoxy-2,5-dimethoxyphenylethanamine series. They have received scant mention in the literature — Shulgin includes only one in *PIHKAL*,^[3] an unenthusiastic account of 2C-O-4.²

Cheng and Castagnoli report shifts between 3.97 and 3.78 ppm for the three methoxy substituents of 2C-O,³ the 4-oxo analogue of 2C-T, a substantial and indicative downfield shift.^[66]

^1H NMR spectra have been reported for 2C-T-2 in CDCl_3 ,^[67] in CD_3OD ,^[43] and in D_2O ^[48,68]; for 2C-T-4 in CDCl_3 ,^[69] and in D_2O ^[48]; for 2C-T-7 in CDCl_3 ,^[69] in CD_3OD ,^[43] and in D_2O .^[48,70]

We have not found ^1H NMR spectra reported for 2C-T. The ^1H NMR spectrum *has* been reported for the α -methyl homologue⁴ of 2C-T in D_2O ,^[48] and for the α -methyl, α -ethyl, and *N*-methyl homologues of the 2C-T series (and the Ψ -2C-T series⁵) in D_2O .^[71] Trachsel has synthesized additional members of the 2C-T series and has reported their spectra in D_2O .^[72]

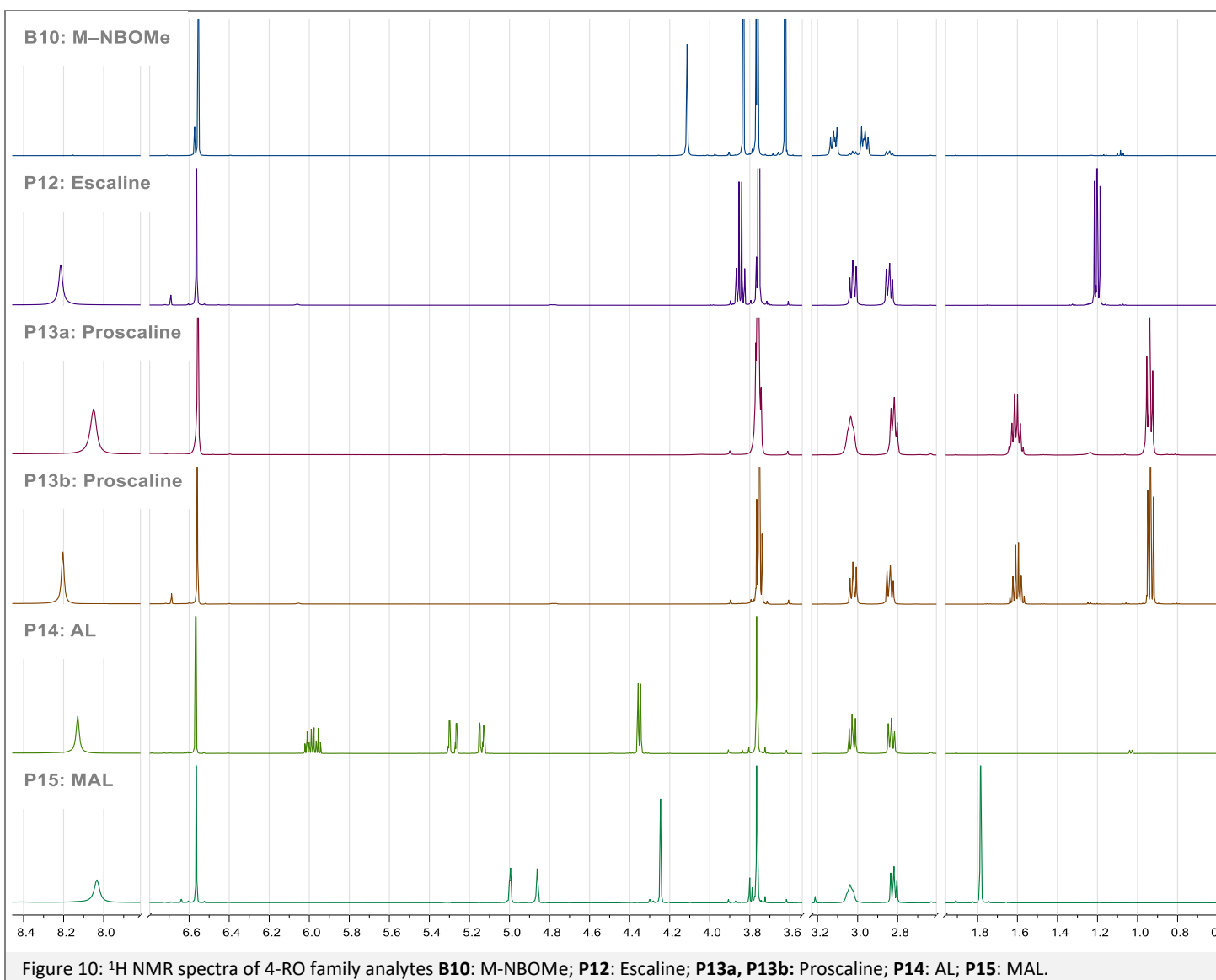
¹ Formerly *thioether*.

² 2C-O-4: 2-[2,5-dimethoxy-4-(propan-2-yloxy)phenyl]ethan-1-amine.

³ 2C-O or 2,4,5-TMPEA: 2-(2,4,5-trimethoxyphenyl)ethan-1-amine.

⁴ ALEPH or DOT: 1-[2,5-dimethoxy-4-(methylsulfanyl)phenyl]propan-2-amine.

⁵ Alexander T. Shulgin introduced the prefix Ψ (Psi) as a notation for the 2,4,6-regioisomer of the corresponding 2,4,5-substituted compound.



The 4-RO family: 4-alk[en]yloxy-3,5-dimethoxyphenyl pattern

Turning from the 2,5- to the 3,5-dimethoxy substitution pattern, the 4-RO family compounds are analogues of the natural product mescaline,¹ sometimes called “scalines.”^[73] Being a controlled substance in Canada, we were unable to include mescaline among our analytes. However, the NBOME analogue of mescaline is not controlled and was obtained instead.

Six of the analytes belong to the 4-RO family, having a *para*-substituent of methoxy (**B10**: M-NBOMe), ethoxy (**P12**: Escaline), propoxy (**P13a**, **P13b**: Proscaline), allyloxy (**P14**: AL), and methallyloxy (**P15**: MAL).²

As noted previously, the symmetrical ring substitution of the 4-RO compounds is evident from the 6H methoxy and 2H aryl proton peaks ca. 3.75 and 6.56 ppm, respectively. Here, as with the 4-R family, *para*-substitution leaves the spectrum otherwise undisturbed.

¹H NMR spectra have been reported for mescaline in DMSO-*d*₆^[29,74] and in CDCl₃^[75–77], for proscaline in DMSO-*d*₆.^[53] We have not found ¹H NMR spectra reported for M-NBOMe, escaline, AL, or MAL.⁵ ¹H NMR spectra have been reported for the α -methyl homologues of proscaline,³ AL,⁴ and MAL⁵ in D₂O.^[78] ¹H NMR spectra have been reported for TMA,⁶ the α -

¹ Mescaline or M: 2-(3,4,5-trimethoxyphenyl)ethan-1-amine.

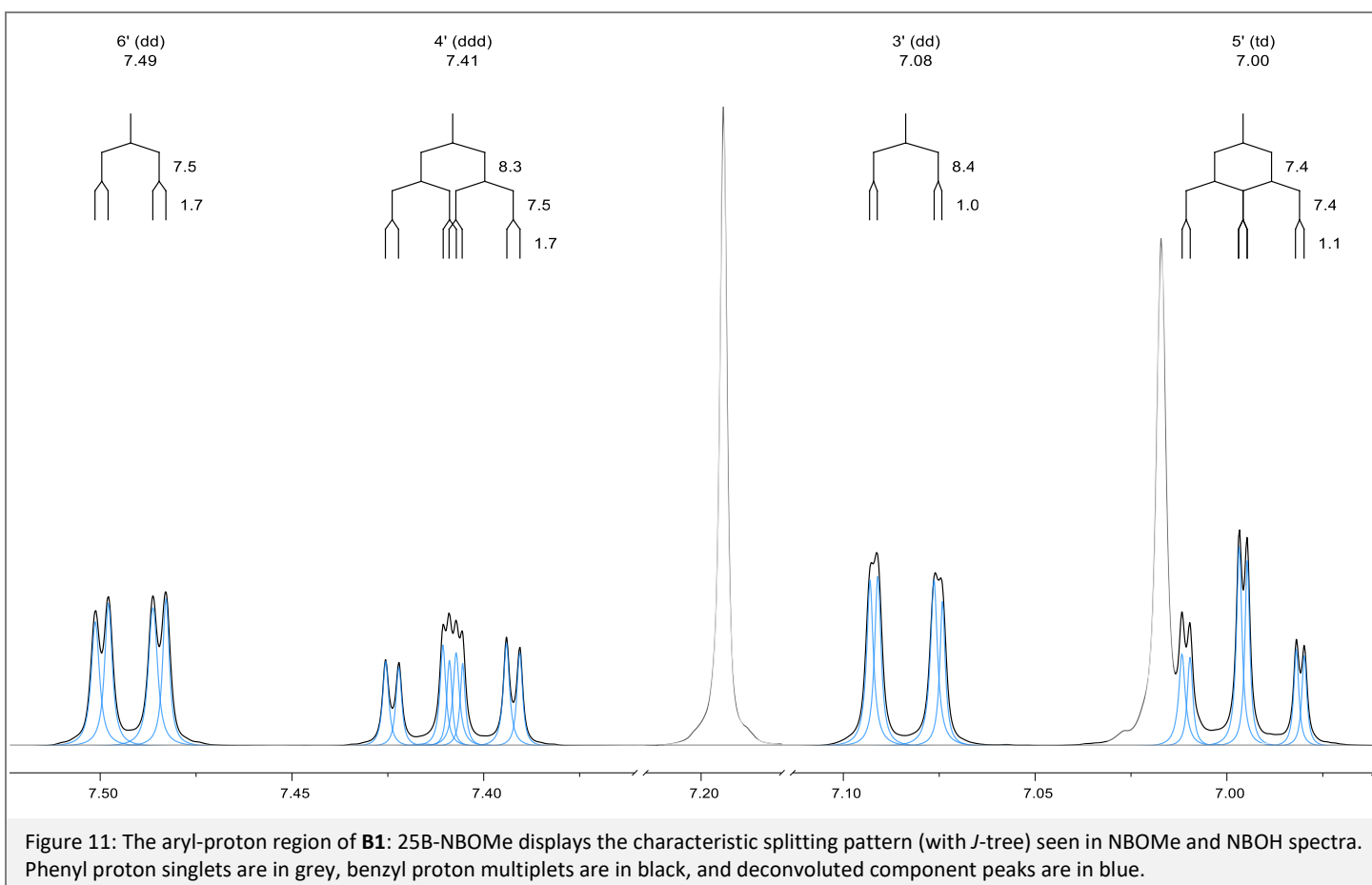
² While *allyloxy* remains acceptable *in general*, prop-2-en-1-yloxy is the PIN. However, *methallyloxy* and *methylallyloxy* are deprecated; (2-methylprop-2-en-1-yl)oxy is the PIN. We’ve opted for brevity over dogma in this case.

³ 3C-P: 1-(3,5-dimethoxy-4-propoxyphenyl)propan-2-amine.

⁴ 3C-AL: 1-[3,5-dimethoxy-4-(prop-2-en-1-yloxy)phenyl]propan-2-amine.

⁵ 3C-MAL: 1-[3,5-dimethoxy-4-[(2-methylprop-2-en-1-yl)oxy]phenyl]propan-2-amine.

⁶ TMA: 1-(3,4,5-trimethoxyphenyl)propan-2-amine.



methyl homologue of mescaline, and for TMA-2¹ and TMA-6,² two regioisomers of interest, in CD₃OD.^[79]

Recently, a regioisomer of mescaline, isomescaline,³ was prepared and the ¹H NMR spectrum in DMSO-*d*₆ reported by Martins.^[30] Reportedly without activity, it seems an unlikely candidate for the grey market.^[3]

The NBOMe and NBOH splitting pattern

The *N*-benzyl protons of the NBOMe and NBOH analytes resonate with a distinctive splitting pattern and at highly consistent shifts of 7.5–7.0 ppm (see Table 4). The NBOMe splitting pattern is consistent with a largely first-order ABCD spin system having six coupling constants: three ortho (³*J*_{3'4'}, ³*J*_{4'5'}, ³*J*_{5'6'} ≈ 8.0 Hz), two meta (⁴*J*_{3'5'}, ⁴*J*_{4'6'} ≈ 2.5 Hz), and one para (⁵*J*_{3'6'} ≈ 0). The aryl region of the **B1**: 25B-NBOMe spectrum and the first-order *J*-tree analysis is shown in Figure 11.

Each of the *outer* two protons (3' and 6') is *ortho*- and *meta*-coupled to the *inner* protons (4' and 5') and appears as a doublet of doublets (dd). The two inner protons are also *ortho*-coupled to each other, producing a further split into a doublet

of doublet of doublets (ddd). Because the *ortho*-coupling constants are quite close — less than 1 Hz apart — a ddd coupling may appear as a triplet of doublets (td). There is no apparent *para*-coupling between the outer protons.

Interestingly, both the shifts and the *ordering* of the benzyl peaks may depend on the solvent. Published NBOMe spectra acquired in DMSO-*d*₆ align quite closely with our own and retain the 6', 4', 3', 5' multiplet sequence.^[37 supp. data] In contrast, for a spectrum of 25C-NBOMe acquired in CDCl₃ the overall splitting pattern is preserved but the ordering is not, becoming 6', 4', 5', 3'.^[51] Multiplets 3' and 5' have traded places.

¹ TMA-2: 1-(2,4,5-trimethoxyphenyl)propan-2-amine.

² TMA-6: 1-(2,4,6-trimethoxyphenyl)propan-2-amine.

³ Isomescaline or IM: 2-(2,3,4-trimethoxyphenyl)ethan-1-amine.

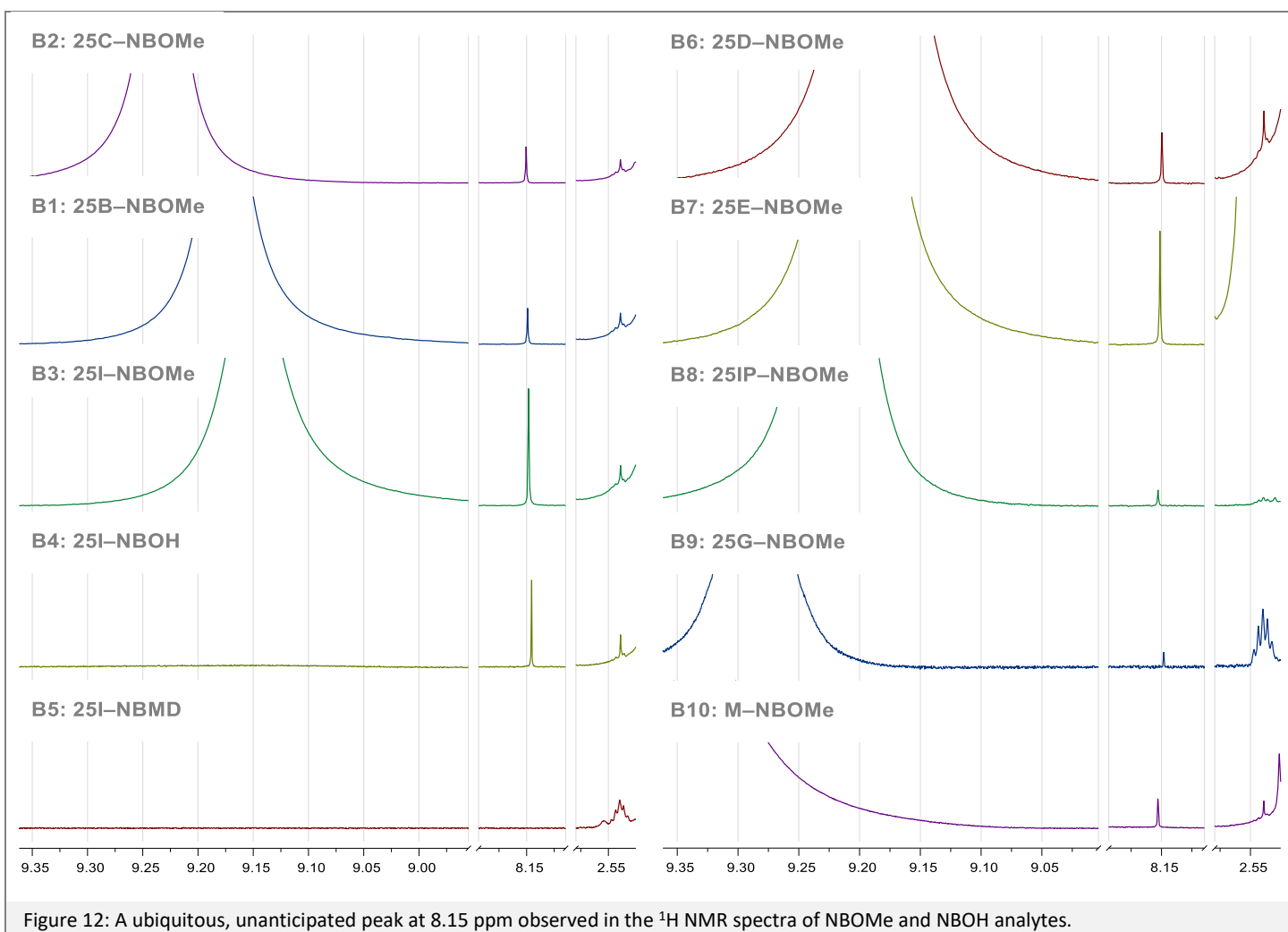


Figure 12: A ubiquitous, unanticipated peak at 8.15 ppm observed in the ^1H NMR spectra of NBOMe and NBOH analytes.

A ubiquitous, unanticipated peak at 8.15 ppm

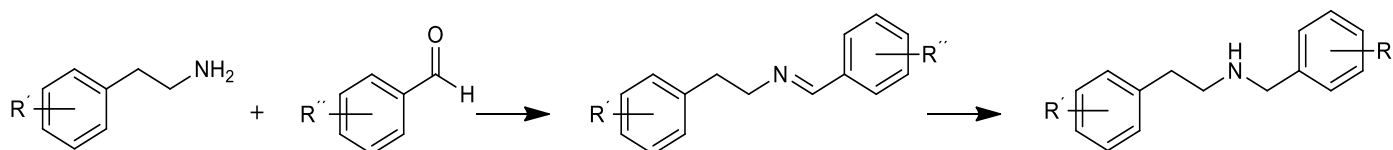
A characteristic peak was noted in the spectrum for 9 of the 10 *N*-benzyl analytes (Figure 12). We initially suspected the sharp singlet at 8.15 ppm was the signal from residual amounts of the benzaldehyde precursors. However, literature shifts were notable more downfield than anticipated, appearing around 10.2–10.3 ppm.^[80,81]

Following the synthetic route outlined by Casale and Hays (Scheme 1), we wondered if the peak could be coming from traces of the unreduced imine intermediate of the reductive amination of the parent phenethylamine with the respective benzaldehyde.^[82] Kappe et al. reported a singlet at 8.15 ppm in the spectrum of the unsubstituted imine (*E*)-1-phenyl-*N*-(2-

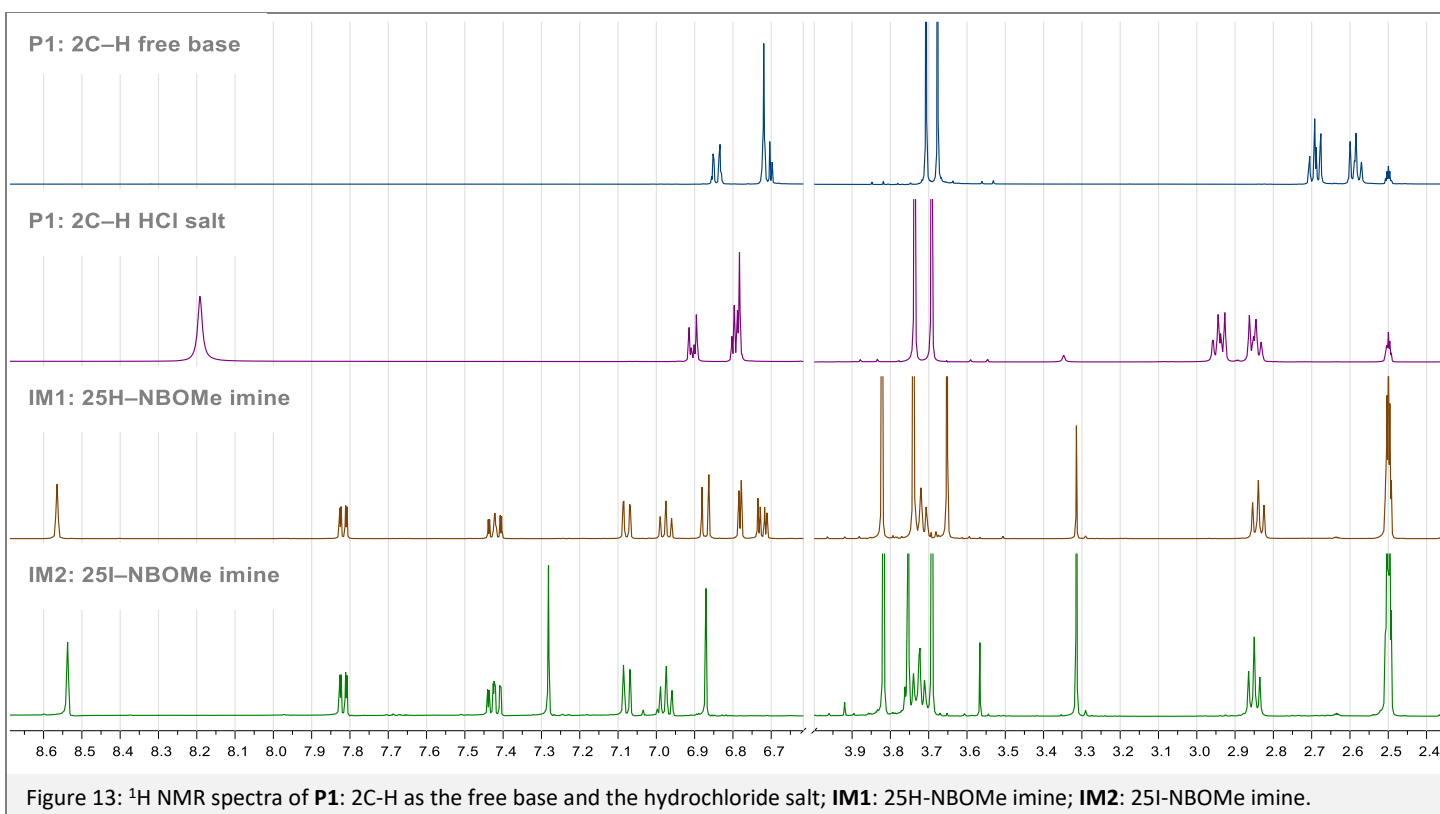
phenylethyl)methanimine (**1**), which seemed promising.^[83]

To test this hypothesis we acquired the ^1H NMR spectra of two commercially available *N*-benzylidene phenylethanamines: **IM1**: 25H-NBOMe imine (**2**) and **IM2**: 25I-NBOMe imine (**3**).

The spectra of the reference did not conclusively match the observed impurities for the respective compounds. The imine analog of 25H-NBOMe, **IM1**, gave a singlet at 8.56 ppm; **IM2**, the imine analog of 25I-NBOMe, a singlet at 8.54 ppm. The distinct shifts from the observed 8.13–8.15 ppm shift made the imine an unlikely candidate. Potential precursors 2-methoxybenzaldehyde, 1,3-benzodioxole-4-carbaldehyde, the unsubstituted primary phenylethanamines, and residual solvents were each investigated but proved unsupported by



Scheme 1: Synthetic route to *N*-benzyl phenylethanamines via the *N*-benzylidene, following Casale and Hays.^[82]



experimental, predicted, or literature ^1H NMR shift values. Though this peak might indeed be an artifact of Scheme 1 we did not pursue this line any further.

The spectrum of **B5**: 25I-NBMD is unique among the *N*-benzyl analytes in having no peak at 8.15 ppm. We believe **B5** is also the only free base *N*-benzyl analyte, having a broad upfield amine peak and no downfield aminium peak. Our analytes include primary, secondary, and tertiary amines; the peak at 8.15 ppm is present only in the spectra of protonated secondary amines.

In each case the peak's shift is precisely 8.15 ppm, while the shift the aminium peaks differ by as much as 0.18 ppm. This seems to rule out the peaks being somehow coupled. No correlation was found between the area of the two peaks, further evidence against coupling.

Approaching the problem from another angle we searched the literature for peaks of shift 8.15 ppm. Robertson et al.^[84] report conditions where the formyl proton of dimethylformamide (DMF) resonates as a sharp singlet at 8.15 ppm in $\text{DMSO}-d_6$.

Though reporting a shift of ca. 7.95 ppm in free DMF, they found the formyl proton signal moves downfield to 8.15 ppm when the amine lone pair forms a hydrogen bond and the quaternary amine orbitals undergo sp^3 hybridization. With the loss of the nitrogen π -orbital, the delocalized bonding found in free DMF is lost (see Figure 15), rotation about the C-N bond is no longer inhibited, and the amine methyl groups become equivalent, resonating as a 6H singlet at 2.54 ppm.

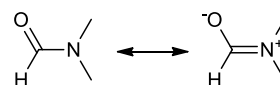


Figure 15: Dimethylformamide resonance structures.

Though the majority of the spectra stacked in Figure 12 indeed include a peak at 2.54 ppm, the protonated DMF theory would seem to demand the area of the upfield dimethyl peak be six times the area of the downfield formyl peak. Tantalizing but far from conclusive. A more satisfactory explanation must await further investigation.

We have not found ^1H NMR spectra reported for 25H-NBOMe

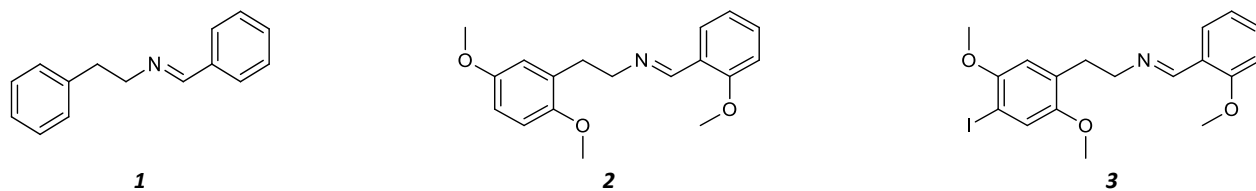


Figure 14: Structure of *N*-benzylidene phenylethanamines

imine or 25I-NBOMe imine. Spectra in CDCl₃ are available from the vendor, Cayman Chemical.

Conclusion

We found only one clear case of significant misrepresentation among our analytes. Analyte **P6a** was sold and labelled as 2C-C, a claim unsupported by the ¹H NMR spectrum which suggests 2C-E. In addition, the spectrum of **P6a** aligns closely with that of **P3**: 2C-E, purchased from the same vendor around the same time. It seems likely analyte **P6a** was mislabelled by the vendor.

Fortunately, because 2C-C and 2C-E are active at comparable levels^[3] this did not constitute a public health emergency as happened in 2009, for example, when the extremely potent and long-acting compound bromo-dragonFLY¹ was sold as 2C-B-FLY,² an order of magnitude less active.^[21,85] Furthermore, the vendor of the mislabelled 2C-E closed several years ago.

On reflection — and the future of open, “legitimate” research

“Experience is essential and to become really proficient in this area, you need to critically examine literally thousands of spectra.”^[86]

When we began this investigation our knowledge of NMR was limited, as was our experience interpreting spectra. The sophisticated and highly regarded Mestrelab software may, in retrospect, have inspired unrealistic expectations that interpreting spectra would be simple, even intuitive. Remarkably, as our investigation progressed so too did our intuition. As a non-traditional, self-directed approach to the study of NMR we found it highly effective. As a potential technique for harm reduction we concede “walk-in” ¹H NMR analysis is unlikely to supplant spot-colour testing any time soon.

The legal status of phenylethanamines in Canada has long been clear.^[36] A terse notice released 1 August 2015 may muddy the water. Health Canada proposes to add “2C-phenethylamines and their salts, derivatives, and isomers, and the salts of their derivatives and isomers” to Schedule III of the Controlled Drugs and Substances Act (CDSA).^[87,88]

What that *means* is far from clear — no elaboration is provided and the CDSA is equally unhelpful, absent any definition for 2C-phenethylamine, salt, derivative, or isomer. The legal status of a substance may be contested when made woolly, even if by design — what fresh *grey area* is this? Unfortunately, the

inevitable consequences of further barriers to open, legitimate research are well documented.^[89–95]

Acknowledgements

The opinions expressed in this article are solely those of the authors and do not necessarily reflect the view of any person, organization, institution, or corporation named in these acknowledgements.

We are hugely indebted to Timothy Burrow, Darcy Burns, and Dmitry Pichugin of the University of Toronto NMR facility for their generous assistance and encouragement. Simon Brandt’s thoughtful insight and unflagging direction has been invaluable. Elias Verweken’s many candid and bracing observations were gratefully received. Peter Ng reviewed the tabular and supplementary data for consistency and offered helpful suggestions. S.J.C. thanks Lee Fadness for his tacit support and generous accommodation, over many years.

We acknowledge the exemplary support and assistance provided by Jedi Masters Carlos Cobas, Santiago Domínguez, Cristina Geadá, Olalla Lema, and Pablo Monje of Mestrelab Research, and the donation of an academic licence. Cayman Chemical provided ¹H NMR CDCl₃ spectra of their *N*-benzylidene products. All procurement and analysis was done in Canada under the auspices of Isomer Design, Toronto.

Finally and with deep regret we mark the loss of distinguished psychedelic researcher and author Alexander “Sasha” Shulgin.

References

- [1] S.J. Chapman, A.A. Avanes. PeakAL: Protons I Have Known and Loved. Supplementary data. Blotter: Absorbing bits of material. Isomer Design. 2015. <http://dx.doi.org/10.16889/isomerdesign-1-sup>
- [2] International Union of Pure and Applied Chemistry (IUPAC), H.A. Favre, W.H. Powell. *Nomenclature of organic chemistry: IUPAC recommendations and preferred names 2013*. Royal Society Of Chemistry, Cambridge, UK, 2014.
- [3] A.T. Shulgin, A. Shulgin. *PiHKAL: A chemical love story*. Transform Press, Berkeley, CA, 1991.
- [4] S.D. Brandt, T. Passie. Research on psychedelic substances. *Drug Test. Analysis* 2012, 4, 539–542. <http://dx.doi.org/10.1002/dta.1389>
- [5] D. Trachsel, C. Enzensperger, D. Lehmann. *Phenethylamine: von der Struktur zur Funktion*. Nachtschatten Science, Solothurn, Switzerland, 2013.
- [6] Multidisciplinary Association for Psychedelic Studies (MAPS). Conference videos. Psychedelic Science 2013, Oakland, CA. Available at: <http://www.maps.org/conference> (accessed 1 Aug 2015)
- [7] L.A. King. New phenethylamines in Europe. *Drug Test. Analysis* 2014, 6, 808–818. <http://dx.doi.org/10.1002/dta.1570>
- [8] D.E. Nichols, W.E. Fantegrossi. Emerging designer drugs, in *The effects of drug abuse on the human nervous system*, Elsevier, 2014, pp. 575–596. <http://dx.doi.org/10.1016/B978-0-12-418679-8.00019-8>
- [9] R.A. Glennon, M. Dukat, M. El-Bermawy, H. Law, J. De Los Angeles, M. Teitler, A. King, K. Herrick-Davis. Influence of amine substituents on 5-HT_{2A} versus 5-HT_{2C} binding of phenylalkyl- and indolylalkylamines. *J. Med. Chem.* 1994, 37, 1929–1935. <http://dx.doi.org/10.1021/jm00039a004>

¹ Bromo-dragonFLY: 1-(8-bromobenzo[1,2-*b*:4,5-*b'*]difuran-4-yl)propan-2-amine.

² 2C-B-FLY: 2-(8-bromo-2,3,6,7-tetrahydrobenzo[1,2-*b*:4,5-*b'*]difuran-4-yl)ethan-1-amine.

- [10] H.H. Pertz, A. Rheineck, S. Elz. N-Benzylated derivatives of the hallucinogenic drugs mescaline and escaline as partial agonists at vascular 5-HT_{2A} receptors. *N-S. Arch. Pharmacol.* **1999**, 359, R29.
- [11] R. Heim, H.H. Pertz, S. Elz. Preparation and *in-vitro* pharmacology of novel secondary amine-type 5-HT_{2A}-receptor agonists: From submillimolar to subnanomolar activity. *Arch. Pharm. Pharm. Med. Chem.* **1999**, 331, 34.
- [12] R. Heim, S. Elz. Novel extremely potent partial 5-HT_{2A}-receptor agonists: Successful application of a new structure-activity concept. *Arch. Pharm. Pharm. Med. Chem.* **2000**, 333, 18.
- [13] H.H. Pertz, R. Heim, S. Elz. N-Benzylated phenylethanamines are highly potent partial agonists at 5-HT_{2A} receptors. *Arch. Pharm. Pharm. Med. Chem.* **2000**, 333, 30.
- [14] R. Heim, H.H. Pertz, S. Elz. Partial 5-HT_{2A}-receptor agonists of the phenylethylamine series: Effect of a trifluoromethyl substituent. *Arch. Pharm. Pharm. Med. Chem.* **2000**, 333, 45.
- [15] R. Heim. Synthesis and pharmacology of potent 5-HT_{2A} receptor agonists which have a partial *N*-2-methoxybenzyl structure, Doctoral thesis, Free University of Berlin: Berlin, Germany, **2003**. Available at: http://www.diss.fu-berlin.de/diss/receive/FUDISS_thesis_000000001221?lang=en (accessed 16 Aug 2014)
- [16] M. Hansen. Design and synthesis of selective serotonin receptor agonists for positron emission tomography imaging of the brain, PhD thesis, University of Copenhagen: Copenhagen, Denmark, **2010**.
- [17] UK Parliament. *The Misuse of Drugs Act 1971 (Temporary Class Drug) Order 2013*, S.I. 2013/1294, 10 Jun 2013. Available at: <http://www.legislation.gov.uk/ukxi/2013/1294/contents/made> (accessed 16 Aug 2014)
- [18] UK Parliament. *The Misuse of Drugs Act 1971 (Ketamine etc.) (Amendment) Order 2014*, S.I. 2014/1106, 10 Jun 2014. Available at: <http://www.legislation.gov.uk/ukxi/2014/1106/contents/made> (accessed 16 Aug 2014)
- [19] Drug Enforcement Administration. *Temporary placement of three synthetic phenethylamines into schedule I*, 78 FR 68716, 15 Nov 2013. Available at: <https://federalregister.gov/a/2013-27315> (accessed 16 Aug 2014)
- [20] S.W. Smith, F.M. Garlich. Availability and supply of novel psychoactive substances, in *Novel psychoactive substances* (Eds: P.I. Wood, D.M. Dargan), Academic Press, Boston, **2013**, pp. 55–77. <http://dx.doi.org/10.1016/B978-0-12-415816-0.00003-1>
- [21] M. Power. *Drugs 2.0: The web revolution that's changing how the world gets high*. Portobello Books, London, **2014**.
- [22] EMCDDA. The Internet and drug markets: Summary of results from an EMCDDA trendspotter study. European Monitoring Centre for Drugs and Drug Addiction. **2015**. Available at: <http://www.emcdda.europa.eu/publications/technical-reports/internet-drug-markets> (accessed 13 Mar 2015)
- [23] EMCDDA. New psychoactive substances in Europe: An update from the EU early warning system. European Monitoring Centre for Drugs and Drug Addiction. **2015**. Available at: <http://www.emcdda.europa.eu/publications/2015/new-psychoactive-substances> (accessed 1 Aug 2015)
- [24] S.D. Brandt, L.A. King, M. Evans-Brown. The new drug phenomenon. *Drug Test. Analysis* **2014**, 6, 587–597. <http://dx.doi.org/10.1002/dta.1686>
- [25] Y.B. Monakhova, T. Kuballa, S. Löbell-Behrends, S. Maixner, M. Kohl-Himmelseher, W. Ruge, D.W. Lachenmeier. Standardless ¹H NMR determination of pharmacologically active substances in dietary supplements and medicines that have been illegally traded over the Internet. *Drug Test. Analysis* **2013**, 5, 400–411. <http://dx.doi.org/10.1002/dta.1367>
- [26] DanceSafe. Drug checking: testing kit instructions. DanceSafe.org. Available at: <https://dancesafe.org/testing-kit-instructions> (accessed 1 Aug 2015)
- [27] RCSources. SafeOrScam definitively gone? Reddit.com. Available at: <http://redd.it/3curzr> (accessed 1 Aug 2015)
- [28] P.A. Hays, J.F. Casale. Characterization of eleven 2,5-dimethoxy-*N*-(2-methoxybenzyl)phenethylamine (NBOMe) derivatives and differentiation from their 3- and 4-methoxybenzyl analogues - Part II. *Microgram J.* **2014**, 11, 3–22. Available at <http://forendex.southernforensic.org/uploads/references/MicrogramJournal/11.1-4.3.22.pdf> (accessed 1 Aug 2015)
- [29] D.J.M. Martins. Analysis of new psychoactive substances: a contribution to forensic chemistry, MSc thesis, Universidade do Porto: Porto, **2014**.
- [30] A.C. Cheng, N. Castagnoli. Synthesis and physicochemical and neurotoxicity studies of 1-(4-substituted-2,5-dihydroxyphenyl)-2-aminoethane analogs of 6-hydroxydopamine. *J. Med. Chem.* **1984**, 27, 513–520. <http://dx.doi.org/10.1021/jm00370a014>
- [31] N.V. Cozzi. Pharmacological studies of some psychoactive phenylalkylamines: entactogens, hallucinogens, and anorectics, PhD thesis, University Of Wisconsin-Madison: Madison, WI, **1994**.
- [32] K. Kanai, K. Takekawa, T. Kumamoto, T. Ishikawa, T. Ohmori. Simultaneous analysis of six phenethylamine-type designer drugs by TLC, LC-MS, and GC-MS. *Forensic Toxicol.* **2008**, 26, 6–12. <http://dx.doi.org/10.1007/s11419-008-0041-2>
- [33] SWGDRUG. 2,5-Dimethoxyphenethylamine (2C-H). Drug monographs. **2014**. Available at: <http://www.swgdrug.org/Monographs/2C-H.pdf> (accessed 12 Feb 2015)
- [34] R.C. Shaler, J.J. Padden. Identification of hallucinogens in illicit seizures I: 2,5-Dimethoxyamphetamine. *J. Pharm. Sci.* **1972**, 61, 1851–1855. <http://dx.doi.org/10.1002/jps.2600611142>
- [35] C. Clark. The identification of methoxy-*N*-methylamphetamines. *J. Forensic Sci.* **1984**, 29, 1056–1071. <http://dx.doi.org/10.1520/JFS11772J>
- [36] Canada Parliament. *Controlled Drugs and Substances Act*, S.C. 1996, c. 19, 20 Jun 1996. Available at: <http://laws-lois.justice.gc.ca/PDF/C-38.8.pdf> (accessed 16 Aug 2014)
- [37] M. Hansen, K. Phonekeo, J.S. Paine, S. Leth-Petersen, M. Begtrup, H. Bräuner-Osborne, J.L. Kristensen. Synthesis and structure-activity relationships of *N*-benzyl phenethylamines as 5-HT_{2A/2C} agonists. *ACS Chem. Neurosci.* **2014**, 5, 243–249. <http://dx.doi.org/10.1021/cn400216u>
- [38] J.D. Power, P. Kavanagh, J. O'Brien, M. Barry, B. Twamley, B. Talbot, G. Dowling, S.D. Brandt. Test purchase, identification and synthesis of 2-amino-1-(4-bromo-2, 5-dimethoxyphenyl)ethan-1-one (bk-2C-B). *Drug Test. Analysis* **2015**, 7, 512–518. <http://dx.doi.org/10.1002/dta.1699>
- [39] Anon. Report on NMR spectroscopy of bk-2CB. bk-2C-B Drug Info. Drugs-Forum. **2013**. Available at: http://isomerdesign.com/doi/1/references/NMR-report-on-BK-2CB_UK.pdf (accessed 6 Apr 2015)
- [40] F.A. Ragan, S.A. Hite, M.S. Samuels, R.E. Garey. 4-Bromo-2,5-dimethoxyphenethylamine: Identification of a New Street Drug. *J. Anal. Toxicol.* **1985**, 9, 91–93. <http://dx.doi.org/10.1093/jat/9.2.91>
- [41] T. Kanamori, H. Inoue, Y. Iwata, Y. Ohmae, T. Kishi. *In vivo* metabolism of 4-bromo-2,5-dimethoxyphenethylamine (2C-B) in the rat: Identification of urinary metabolites. *J. Anal. Toxicol.* **2002**, 26, 61–66. <http://dx.doi.org/10.1093/jat/26.2.61>
- [42] SWGDRUG. 4-Bromo-2,5-dimethoxyphenethylamine (2C-B). Drug monographs. **2005**. Available at: <http://www.swgdrug.org/Monographs/4-bromo-2,5-dimethoxyphenethylamine.pdf> (accessed 12 Feb 2015)
- [43] M. Takahashi, M. Nagashima, J. Suzuki, T. Seto, I. Yasuda, T. Yoshida. Analysis of phenethylamines and tryptamines in designer drugs using gas chromatography-mass spectrometry. *J. Health Sci.* **2008**, 54, 89–96. <http://dx.doi.org/10.1248/jhs.54.89>
- [44] C. Giroud, M. Augsburger, L. Rivier, P. Mangin, F. Sadeghipour, E. Varesio, J.L. Veuthey, P. Kamalaprjia. 2C-B: A new psychoactive phenylethylamine recently discovered in ecstasy tablets sold on the Swiss black market. *J. Anal. Toxicol.* **1998**, 22, 345–354. <http://dx.doi.org/10.1093/jat/22.5.345>
- [45] T. Ando, M. Sugiyama, N. Osaki, J. Watase. Analysis of phenethylamine analogues. Central Customs Laboratory, Ministry of Finance, Japan. **2011**. Available at: http://www.customs.go.jp/ccl_search/e_info_search/drugs/r_51_07_e.pdf (accessed 1 Aug 2015)

- [46] SWGDRUG. 2,5-Dimethoxy-4-chlorophenethylamine (2C-C). Drug monographs. **2014**. Available at: <http://www.swgdrug.org/Monographs/2C-C.pdf> (accessed 12 Feb 2015)
- [47] SWGDRUG. 2,5-Dimethoxy-4-iodophenethylamine (2C-I). Drug monographs. **2005**. Available at: <http://www.swgdrug.org/Monographs/2,5-dimethoxy-4-iodophenethylamine.pdf> (accessed 12 Feb 2015)
- [48] M. Ogino, T. Naiki, H. Orui, K. Kosone, M. Yamazaki. Study of method for identifying phenethylamine drugs. Central Customs Laboratory, Ministry of Finance, Japan. no date. Available at: http://www.customs.go.jp/ccl_search/e_info_search/drugs/r_50_08_e.pdf (accessed 17 Aug 2014)
- [49] R.A. Glennon, M.L. Bondarev, N. Khorana, R. Young, J.A. May, M.R. Hellberg, M.A. McLaughlin, N.A. Sharif. B-Oxygenated analogues of the 5-HT_{2A} serotonin receptor agonist 1-(4-bromo-2,5-dimethoxyphenyl)-2-aminopropane. *J. Med. Chem.* **2004**, *47*, 6034–6041. <http://dx.doi.org/10.1021/jm040082s>
- [50] SWGDRUG. 25B-NBOMe. Drug monographs. **2013**. Available at: <http://www.swgdrug.org/Monographs/25B-NBOMe.pdf> (accessed 12 Feb 2015)
- [51] D. Zuba, K. Sekula, A. Buczek. 25C-NBOMe – New potent hallucinogenic substance identified on the drug market. *Forensic Sci. Int.* **2013**, *227*, 7–14. <http://dx.doi.org/10.1016/j.forsciint.2012.08.027>
- [52] SWGDRUG. 25C-NBOMe. Drug monographs. **2012**. Available at: <http://www.swgdrug.org/Monographs/25C-NBOMe.pdf> (accessed 12 Feb 2015)
- [53] G. Serpelloni, T. Macchia, C. Locatelli, C. Rimondo, C. Seri. 25I-NBOMe. New drugs-Nuove sostanze psicoattive. Dipartimento Politiche Antidroga. **2013**. Available at: <http://isomerdesign.com/Cdsa/IT/3.3/843-854.pdf> (accessed 12 Feb 2015)
- [54] SWGDRUG. 25I-NBOMe. Drug monographs. **2014**. Available at: <http://www.swgdrug.org/Monographs/25I-NBOMe.pdf> (accessed 12 Feb 2015)
- [55] D.E. Nichols, S.P. Frescas, B.R. Chemel, K.S. Rehder, D. Zhong, A.H. Lewin. High specific activity tritium-labeled *N*-(2-methoxybenzyl)-2,5-dimethoxy-4-iodophenethylamine (INBMeO): A high-affinity 5-HT_{2A} receptor-selective agonist radioligand. *Bioorg. Med. Chem.* **2008**, *16*, 6116–6123. <http://dx.doi.org/10.1016/j.bmc.2008.04.050>
- [56] D.E. Nichols. 25I-NBOMe ¹H NMR spectrum. Forendex. Southern Association of Forensic Scientists (SAFS). **2012**. Available at: http://forendex.southernforensic.org/uploads/spectra/25I-NBOMe_NMR.pdf (accessed 17 Aug 2014)
- [57] D.E. Nichols, M.F. Sassano, A.L. Halberstadt, L.M. Klein, S.D. Brandt, S.P. Elliott, W.J. Fiedler. *N*-Benzyl-5-methoxytryptamines as potent serotonin 5-HT₂ receptor family agonists and comparison with a series of phenethylamine analogues. *ACS Chem. Neurosci.* **2015**, *6*, 1165–1175. <http://dx.doi.org/10.1021/cn500292d>
- [58] E. Pretsch. *Structure determination of organic compounds: tables of spectral data*. Springer, Berlin, **2009**. <http://dx.doi.org/10.1007/978-3-540-93810-1>
- [59] R.A. Allred. Spectral characterization of 2,4-dimethoxy-3-methylphenethylamine and comparison to 2,5-dimethoxy-4-methylphenethylamine (“2C-D”). *Microgram J.* **2005**, *3*, 16–26. Available at: <http://forendex.southernforensic.org/uploads/references/MicrogramJournal/3.1-2.16.26.pdf> (accessed 1 Aug 2015)
- [60] D. Zuba, K. Sekula. Identification and characterization of 2,5-dimethoxy-3,4-dimethyl-β-phenethylamine (2C-G) – A new designer drug. *Drug Test. Analysis* **2013**, *5*, 549–559. <http://dx.doi.org/10.1002/dta.1396>
- [61] SWGDRUG. 2,5-Dimethoxy-4-ethylphenethylamine (2C-E). Drug monographs. **2014**. Available at: <http://www.swgdrug.org/Monographs/2C-E.pdf> (accessed 12 Feb 2015)
- [62] G. Boatto, M.A. Pirisi, L. Burrai, E. Baralla, M.P. Demontis, M.V. Varoni, M. Nieddu. An LC–MS–MS method for quantitation of four new phenethylamines (BOX series) in plasma: in vivo application. *Forensic Toxicol.* **2014**, *32*, 75–81. <http://dx.doi.org/10.1007/s11419-013-0204-7>
- [63] M.A. Torres, B. Cassels, M.C. Rezende. The Preparation of Potentially Psychoactive β-Alkoxyphenethylamines. *Synth. Commun.* **1995**, *25*, 1239–1247. <http://dx.doi.org/10.1080/00397919508012687>
- [64] D. Zuba, K. Sekula. Analytical characterization of three hallucinogenic *N*-(2-methoxy)benzyl derivatives of the 2C-series of phenethylamine drugs. *Drug Test. Analysis* **2013**, *5*, 634–645. <http://dx.doi.org/10.1002/dta.1397>
- [65] SWGDRUG. 25E-NBOMe. Drug monographs. **2013**. Available at: <http://www.swgdrug.org/Monographs/25E-NBOMe.pdf> (accessed 12 Feb 2015)
- [66] A.C. Cheng, N. Castagnoli. Synthesis and physicochemical and neurotoxicity studies of 1-(4-substituted-2,5-dihydroxyphenyl)-2-aminoethane analogs of 6-hydroxydopamine. *J. Med. Chem.* **1984**, *27*, 513–520. <http://dx.doi.org/10.1021/jm00370a014>
- [67] N. Machiko, S. Takako, M. Takahashi, H. Suzuki, I. Yasuda. Spectrum Data of the 3rd Governor-designated Drugs and the Analyses of Uncontrolled Drugs Purchased Apr. 2005–Mar. 2006. Tokyo Metropolitan Institute of Public Health. **2006**. Available at: <http://www.tokyo-eiken.go.jp/assets/issue/journal/2006/rep.html>
- [68] SWGDRUG. 2,5-Dimethoxy-4-ethylthiophenylamine (2C-T-2). Drug monographs. **2005**. Available at: <http://www.swgdrug.org/Monographs/2,5-dimethoxy-4-ethylthiophenylamine.pdf> (accessed 12 Feb 2015)
- [69] K. Doi, M. Miyazawa, H. Fujii, T. Kojima. Analysis of the chemical drugs among structural isomer. *Yakugaku Zasshi* **2006**, *126*, 815–823. <http://dx.doi.org/10.1248/yakushi.126.815>
- [70] SWGDRUG. 2,5-Dimethoxy-4-n-propylthiophenethylamine (2C-T-7). Drug monographs. **2014**. Available at: <http://www.swgdrug.org/Monographs/2,5-dimethoxy-4-n-propylthiophenethylamine.pdf> (accessed 12 Feb 2015)
- [71] A. Gallardo-Godoy, A. Fierro, T.H. McLean, M. Castillo, B.K. Cassels, M. Reyes-Parada, D.E. Nichols. Sulfur-substituted α-alkyl phenethylamines as selective and reversible MAO-A inhibitors: biological activities, CoMFA analysis, and active site modeling. *J. Med. Chem.* **2005**, *48*, 2407–2419. <http://dx.doi.org/10.1021/jm0493109>
- [72] D. Trachsel. Synthesis of novel (phenylalkyl)amines for the investigation of structure-activity relationships, Part 3. *Helv. Chim. Acta* **2003**, *86*, 2754–2759. <http://dx.doi.org/10.1002/hlca.200390224>
- [73] B. Meyers-Riggs. Leminger’s scalines. countyourculture. Rational exploration of the underground. Available at: <http://countyourculture.com/2012/05/04/leminger-allylescaline> (accessed 1 Aug 2015)
- [74] M. Schulze. Synthesis of 2-arylethylamines by the Curtius rearrangement. *Synth. Commun.* **2010**, *40*, 1461–1476. <http://dx.doi.org/10.1080/00397910903097302>
- [75] M. Kohno, S. Sasao, S.-I. Murahashi. Synthesis of phenethylamines by hydrogenation of β-nitrostyrenes. *Bull. Chem. Soc. Jpn.* **1990**, *63*, 1252–1254. <http://dx.doi.org/10.1246/bcsj.63.1252>
- [76] J.A. Sintas, A.A. Vitale. Synthesis of derivatives of [1-131] phenylalkylamines for brain mapping. *J. Labelled Compd. Radiopharm.* **1998**, *41*, 53–61. [http://dx.doi.org/10.1002/\(SICI\)1099-1344\(199801\)41:1<53::AID-JLCR53>3.0.CO;2-K](http://dx.doi.org/10.1002/(SICI)1099-1344(199801)41:1<53::AID-JLCR53>3.0.CO;2-K)
- [77] E. Alacid, C. Nájera. Regioselective Heck Reaction of *N*-Vinylphthalimide: A General Strategy for the Synthesis of (*E*)-*N*-Styrylphthalimides and Phenethylamines. *Adv. Synth. Catal.* **2008**, *350*, 1316–1322. <http://dx.doi.org/10.1002/adsc.200800074>
- [78] D. Trachsel. Synthese von neuen (phenylalkyl)aminen zur untersuchung von struktur-aktivitätsbeziehungen, Mitteilung 1, Mescaline derivative. *Helv. Chim. Acta* **2002**, *85*, 3019–3026. [http://dx.doi.org/10.1002/1522-2675\(200209\)85:9<3019::AID-HLCA3019>3.0.CO;2-4](http://dx.doi.org/10.1002/1522-2675(200209)85:9<3019::AID-HLCA3019>3.0.CO;2-4)
- [79] K. Tsujikawa, T. Kanamori, K. Kuwayama, H. Miyaguchi, Y. Iwata, H. Inoue. Analytical Profiles for 3,4,5-, 2,4,5-, and 2,4,6-Trimethoxyamphetamine. *Microgram J.* **2006**, *4*, 12–23. Available at: <http://forendex.southernforensic.org/uploads/references/MicrogramJournal/4.1-4.12.23.pdf> (accessed 1 Aug 2015)
- [80] R.J. Abraham, M. Mobli, R.J. Smith. ¹H chemical shifts in NMR: Part 19. Carbonyl anisotropies and steric effects in aromatic aldehydes and ketones. *Magn. Reson. Chem.* **2003**, *41*, 26–36. <http://dx.doi.org/10.1002/mrc.1125>

- [81] A.M. Deveau, T.L. Macdonald. Practical synthesis of biaryl colchicinoids containing 3',4'-catechol ether-based A-rings via Suzuki cross-coupling with ligandless palladium in water. *Tetrahedron Lett.* **2004**, *45*, 803–807. <http://dx.doi.org/10.1016/j.tetlet.2003.11.016>
- [82] J.F. Casale, P.A. Hays. Characterization of eleven 2,5-dimethoxy-N-(2-methoxybenzyl)phenethylamine (NBOMe) derivatives and differentiation from their 3- and 4-methoxybenzyl analogues - Part I. *Microgram Journal* **2012**, *9*, 84–109. Available at <http://forendex.southernforensic.org/uploads/references/MicrogramJournal/11.1-4.3.22.pdf> (accessed 1 Aug 2015)
- [83] T. Kappe, S. Ajili, W. Stadlbauer. Aktive malonester als synthons für heterocyclen: Eine methode zur herstellung von 4-hydroxy-2(1H)-pyridonen. *J. Heterocycl. Chem.* **1988**, *25*, 463–468. <http://dx.doi.org/10.1002/jhet.5570250221>
- [84] G.P. Robertson, S.D. Mikhailenko, K. Wang, P. Xing, M.D. Guiver, S. Kaliaguine. Casting solvent interactions with sulfonated poly(ether ether ketone) during proton exchange membrane fabrication. *J. Membr. Sci.* **2003**, *219*, 113–121. [http://dx.doi.org/10.1016/S0376-7388\(03\)00193-5](http://dx.doi.org/10.1016/S0376-7388(03)00193-5)
- [85] A. Zawadzki. Bromo dragonfly being sold as 2CB-Fly. [Bluelight.org](http://www.bluelight.org). 22 September 2014. Available at: <http://www.bluelight.org/vb/threads/468561-Warning-Bromo-Dragonfly-being-sold-as-2CB-Fly> (accessed 22 Sep 2014)
- [86] S.A. Richards, J.C. Hollerton. *Essential practical NMR for organic chemistry*. John Wiley, Chichester, West Sussex, U.K, **2011**.
- [87] Health Canada. Proposal regarding the scheduling of 2C-phenethylamines under the Controlled Drugs and Substances Act and its regulations. Available at: <http://www.hc-sc.gc.ca/hc-ps/consult/phenethylamines-eng.php> (accessed 1 Aug 2015)
- [88] Health Canada. Proposed order amending Schedule III to the Controlled Drugs and Substances Act to add 2C-phenethylamines and their salts, derivatives, and isomers and the salts of their derivatives and isomers. *C. Gaz. I* **2015**, *149*, 2158–2159. Available at: <http://gazette.gc.ca/rp-pr/p1/2015/2015-01-31/html/notice-avis-eng.php#na1> (accessed 1 Aug 2015)
- [89] A.T. Shulgin, A. Shulgin. Barriers to Research, in *TiHKAL: The continuation*, Transform Press, Berkeley, CA, **1997**, pp. 346–358.
- [90] J. Wiley, J. Marusich, J.W. Huffman, R.L. Balster, B. Thomas. Hijacking of Basic Research: The Case of Synthetic Cannabinoids. RTI Press. **2011**. Available at: <http://www.rti.org/publications/rtipress.cfm?pubid=17971> (accessed 1 Aug 2015)
- [91] D.J. Nutt, L.A. King, D.E. Nichols. Effects of Schedule I drug laws on neuroscience research and treatment innovation. *Nat. Rev. Neurosci.* **2013**, *14*, 577–585. <http://dx.doi.org/10.1038/nrn3530>
- [92] M.P. Monaghan. *Evidence versus politics: exploiting research in UK drug policy making*. Policy, Bristol ; Portland, Ore, **2011**.
- [93] L.A. King, D.J. Nutt, N. Singleton, R. Howard. Analogue controls: an imperfect law. UK Drug Policy Commission, Independent Scientific Committee on Drugs. **2012**. Available at: <http://www.ukdpc.org.uk/wp-content/uploads/Analogue-control-19.06.12.pdf> (accessed 1 Aug 2015)
- [94] F. Measham. Social issues in the use of novel psychoactive substances: differentiated demand and ideological supply, in *Novel Psychoactive Substances* (Eds: P.I. Dargan, D.M. Wood), Academic Press, Boston, **2013**, pp. 105–127.
- [95] N.L. Ginsberg. The politics of research: Science's role in ending the drug war. MAPS. Available at: <http://www.maps.org/bulletin-items/387-bulletin-spring-2015/5668-the-politics-of-research-science%E2%80%99s-role-in-ending-the-drug-war> (accessed 1 Aug 2015)